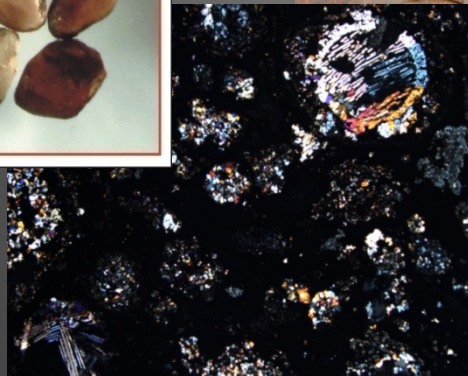


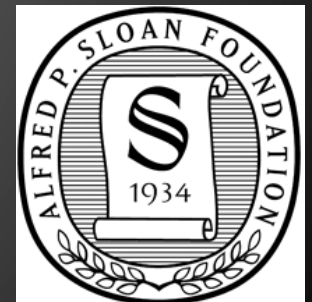
THE STORY OF EARTH: How the Geosphere and Biosphere Co-Evolved



Scientific American—Bright Horizons 17
Robert M. Hazen—11 July 2013



DEEP CARBON OBSERVATORY
deepcarbon.net



CARNEGIE
INSTITUTION FOR
SCIENCE

Mineral Evolution Collaborators

Johns Hopkins Univ.

Dimitri Sverjensky

John Ferry

Namhey Lee

David Azzolini

Univ. of Arizona

Robert Downs

Hexiong Yang

Joaquin Ruiz

Joshua Golden

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Dominic Papineau

Univ. of Maine

Edward Grew

Indiana University

David Bish

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Rodney Ewing

Univ. of Maryland

James Farquhar

Xiaoming Liu

Univ. of Wisconsin

John Valley

Geol. Survey Canada

Wouter Bleeker

CalTech

Ralph Milliken

Smithsonian Inst.

Timothy McCoy

Harvard University

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Mineral Evolution: Outline

- 1. What is mineral evolution?**
- 2. Ten stages of mineral evolution.**
- 3. Implications of mineral evolution.**
- 4. Recent discoveries in mineral evolution.**

What Is Mineral Evolution?

A change over time in:

- The diversity of mineral species
- The relative abundances of minerals
- The compositional ranges of minerals
- The grain sizes and shapes of minerals

Hazen et al. (2008) *Amer. Mineral.* 93, 1693; Hazen et al. (2009) *Amer. Mineral.* 94, 1293;
Hazen et al. (2010) *Elements* 6, #1, 9-46; Hazen et al. (2011) *Amer. Mineral.* 96, 953.

What Is Mineral Evolution?

Focus exclusively on near-surface (<3 km depth) phases.

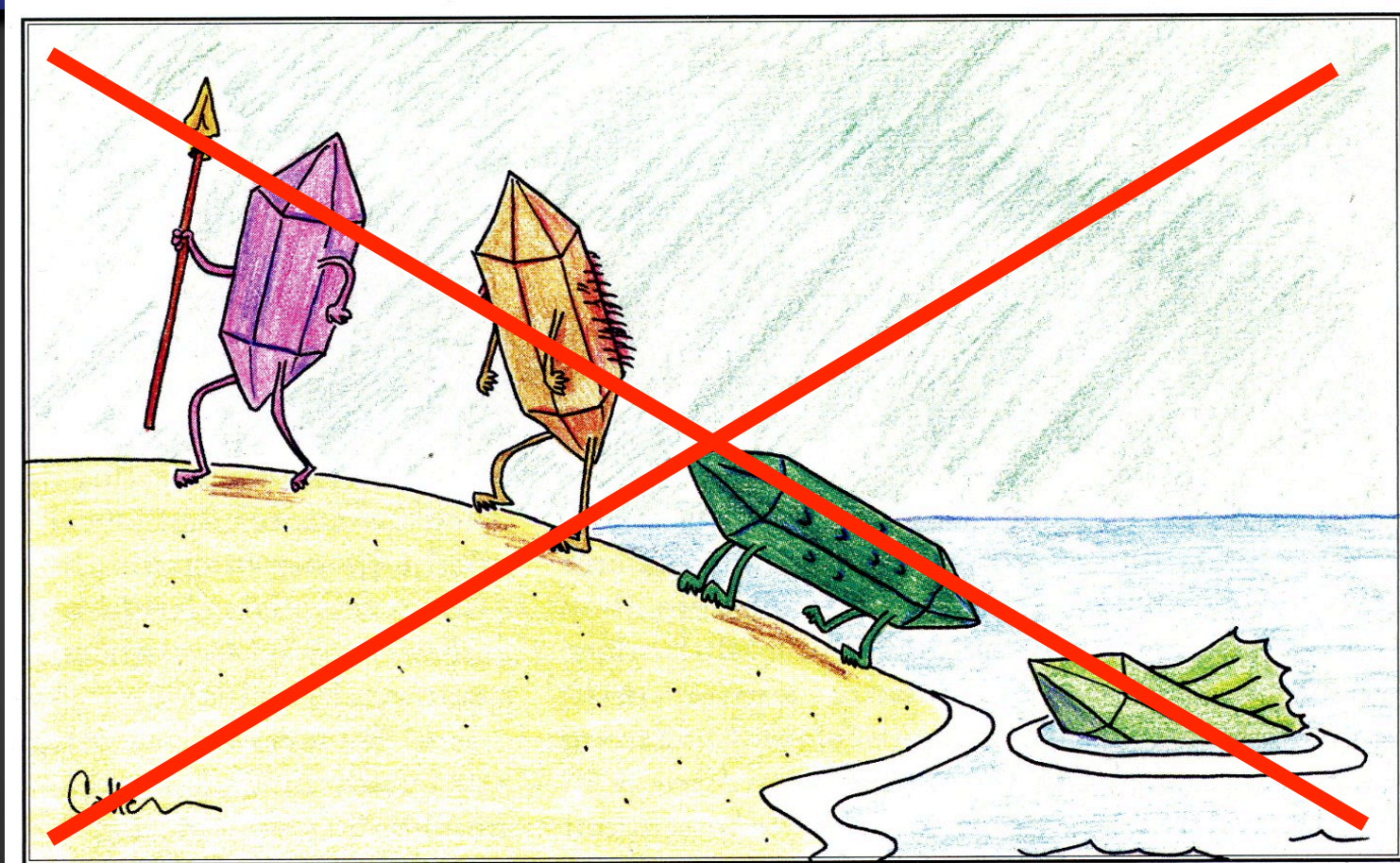
- Accessible to study on Earth
- Most likely to be observed on other planets and moons
- Direct interaction with biology

Hazen et al. (2008) *Amer. Mineral.* 93, 1693; Hazen et al. (2009) *Amer. Mineral.* 94, 1293;
Hazen et al. (2010) *Elements* 6, #1, 9-46; Hazen et al. (2011) *Amer. Mineral.* 96, 953.

Why Mineral Evolution?

- Reframe mineralogy in a dynamic historical context
- Classify terrestrial planets and moons & identify mineralogical targets
- Explore general principles related to complex evolving systems
- Pose new mineralogical questions

A Comment on “Evolution”

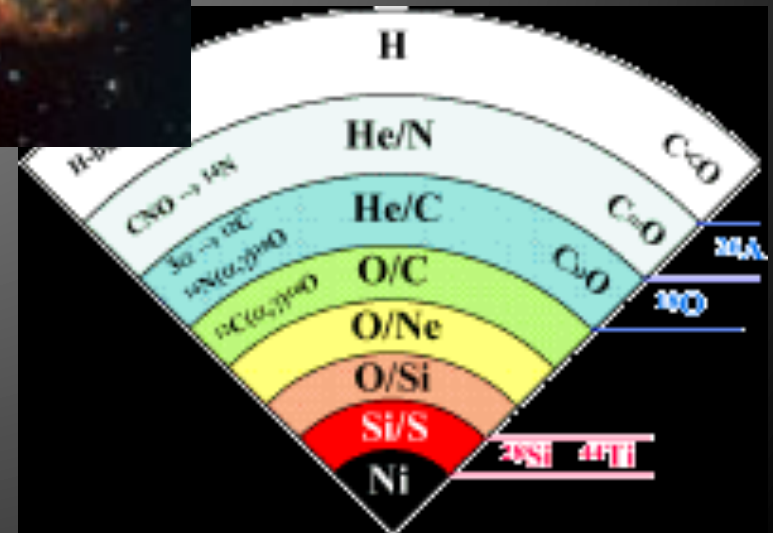
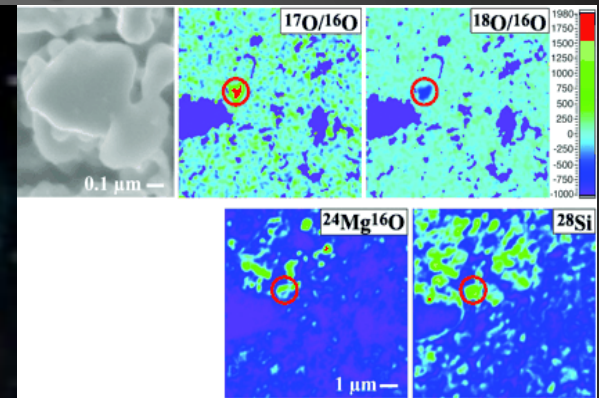


- But NOT Darwinian evolution!

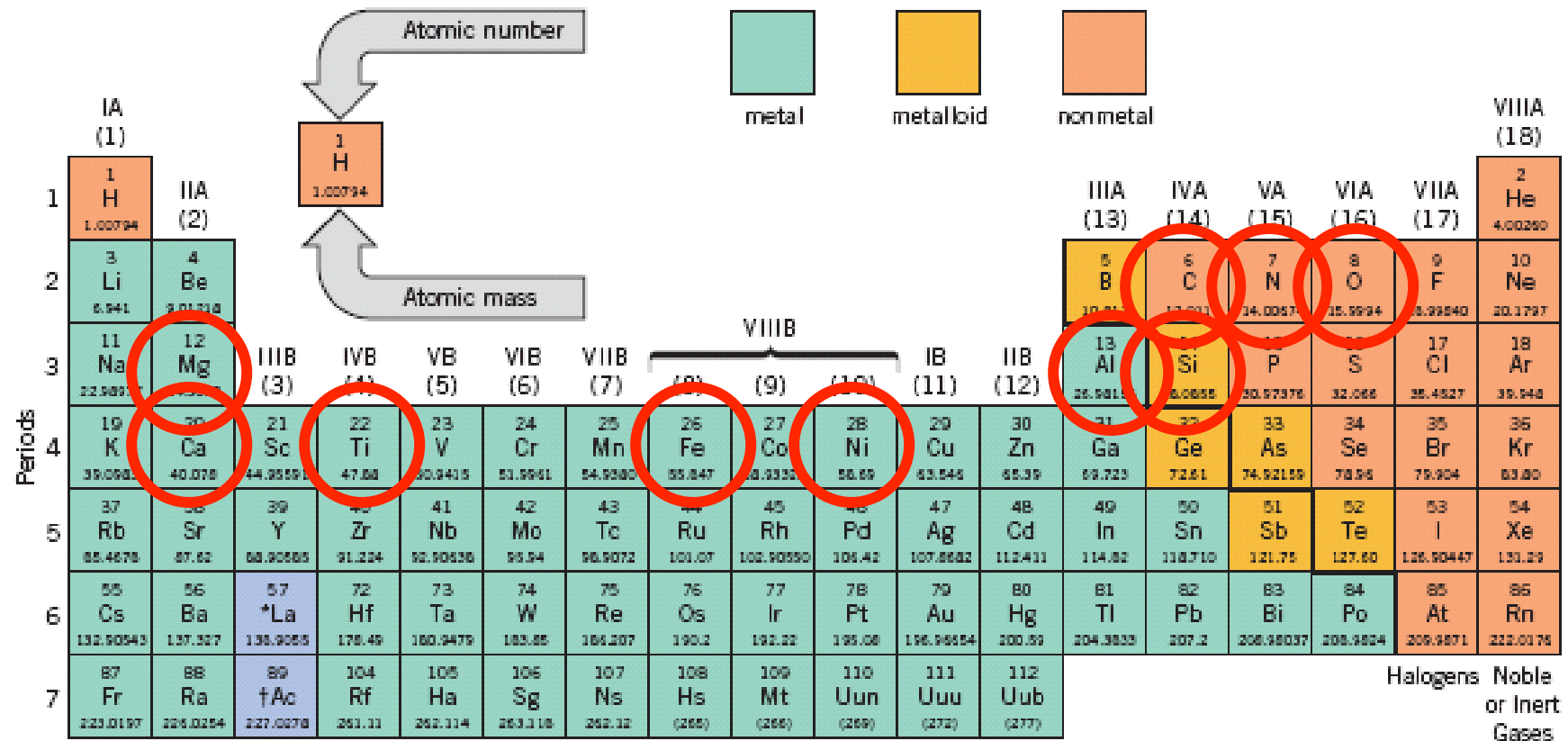
“Ur”-Mineralogy

Pre-solar grains contain about a dozen micro- and nano-mineral phases:

- Diamond/Lonsdaleite
- Graphite (C)
- Moissanite (SiC)
- Osbornite (TiN)
- Nierite (Si_3N_4)
- Rutile (TiO_2)
- Corundum (Al_2O_3)
- Spinel (MgAl_2O_4)
- Hibbonite ($\text{CaAl}_{12}\text{O}_{19}$)
- Forsterite (Mg_2SiO_4)
- Nano-particles of TiC, ZrC, MoC, FeC, Fe-Ni metal within graphite.
- GEMS (silicate glass with embedded metal and sulfide).



Ur-Minerals



14 The periodic table of

+	58 Ce 140.116	59 Pr 140.90768	60 Nd 144.24	61 Pm 144.9127	62 Sm 150.36	63 Eu 151.965	64 Gd 157.25	65 Tb 158.92534	66 Dy 162.50	67 Ho 164.93032	68 Er 167.26	69 Tm 168.93421	70 Yb 173.04	71 Lu 174.967
†	90 Th 232.0381	91 Pa 231.0362	92 U 238.0289	93 Np 237.0482	94 Pu 244.0642	95 Am 243.0614	96 Cm 247.0703	97 Bk 247.0703	98 Cf 247.0703	99 Es 252.083	100 Fm 257.0851	101 Md 258.10	102 No 259.1009	103 Lr 260.105

Mineral Evolution:

How did we get from a dozen minerals (with 10 elements) to >4800 on Earth today?

What does the distribution of minerals through time tell us about key events in Earth history?

What Drives Mineral Evolution?

Deterministic and stochastic processes that occur on any terrestrial body:

1. The progressive separation and concentration of chemical elements from their original uniform distribution.

What Drives Mineral Evolution?

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1. The progressive separation and concentration of chemical elements from their original uniform distribution.
2. An increase in the range of intensive variables (T, P, activities of volatiles).

What Drives Mineral Evolution?

Deterministic and stochastic processes that occur on any terrestrial body:

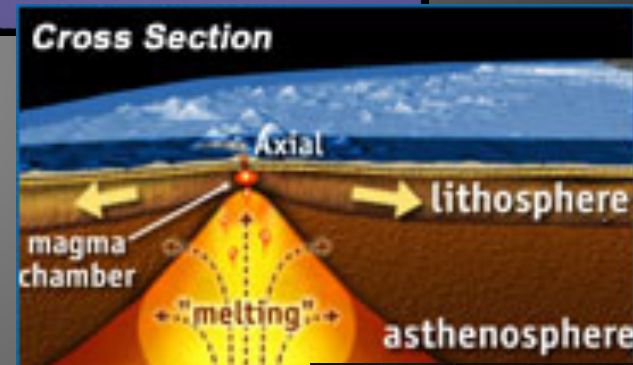
1. The progressive separation and concentration of chemical elements from their original uniform distribution.
2. An increase in the range of intensive variables (T, P, activities of volatiles).
3. **The generation of far-from-equilibrium conditions by living systems.**

Three Eras of Earth's Mineral Evolution

1. The Era of Planetary Accretion



2. The Era of Crust and Mantle Reworking



3. The Era of Bio-Mediated Mineralogy

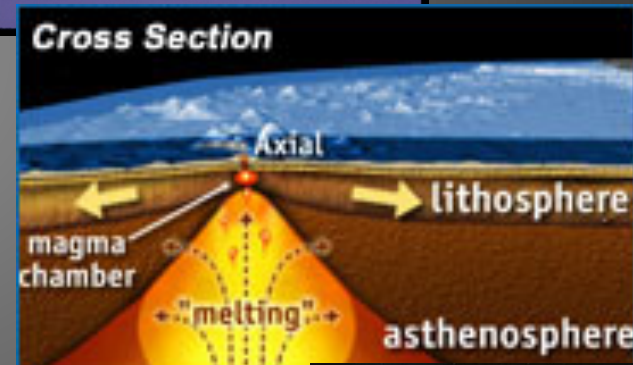


Three Eras of Earth's Mineral Evolution

1. The Era of Planetary Accretion



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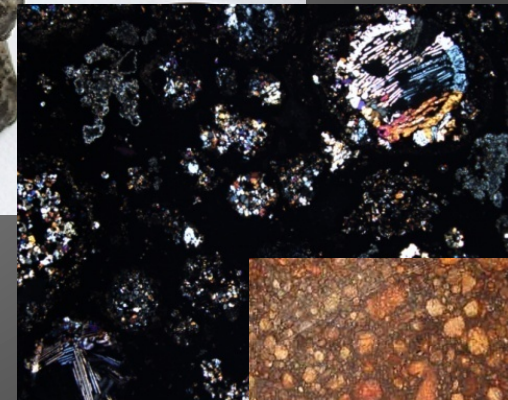


Stage 1: Primary Chondrite Minerals

Minerals formed ~4.56 Ga in the Solar nebula as a consequence of condensation, melt solidification, or recrystallization

~60 mineral species

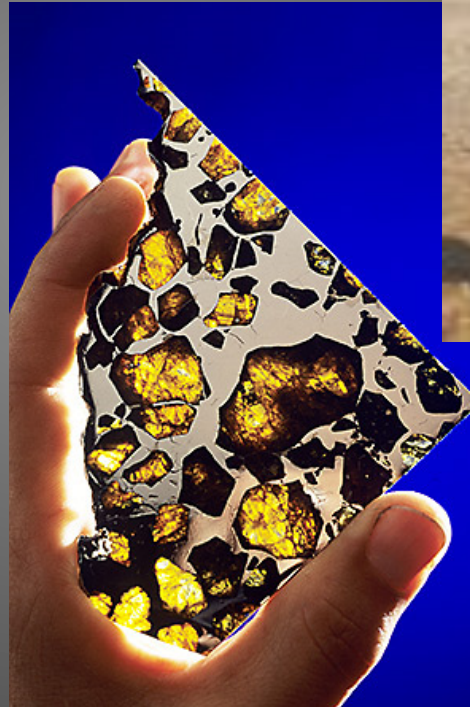
- CAIs
- Chondrules
- Silicate matrix
- Opaque phases



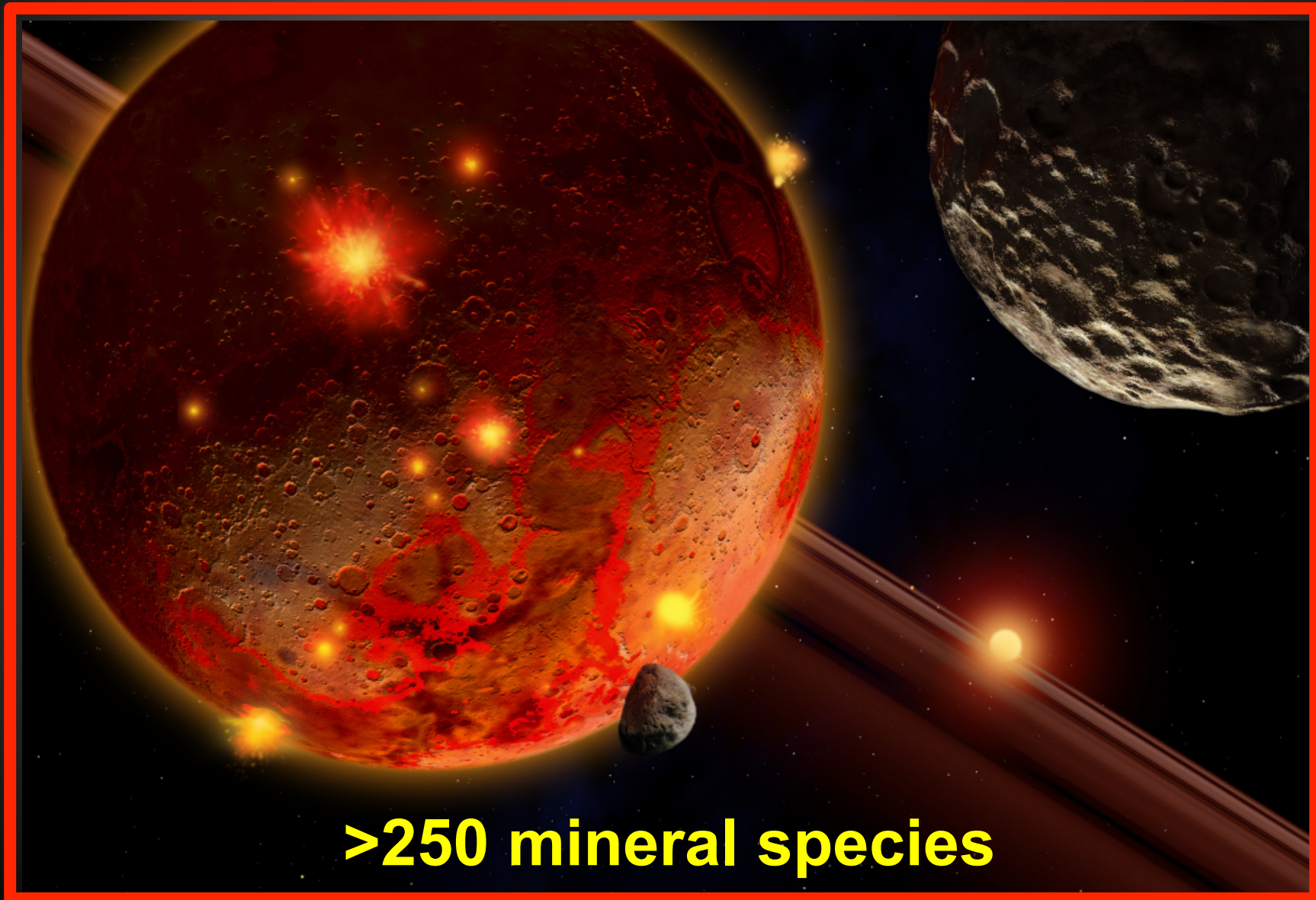
Stage 2: Aqueous alteration, metamorphism and differentiation of planetesimals

~250 mineral known species: 4.56-4.55 Ga

- First albite & K-spar
- First significant SiO_2
- Feldspathoids
- Hydrous biopyriboles
- Clay minerals
- Zircon
- Shock phases



Stage 2: Planetary Accretion



>250 mineral species

Stages 1 and 2: Planetary Accretion

In these early stages all of Earth's near-surface compositional complexity was present, but it was not manifest in a diversity of unusual mineral species.

>250 mineral species

Stages 1 & 2 Minerals

Atomic number

Atomic mass

metal metalloid nonmetal

	IA (1)																	VIIIA (18)
1	1 H 1.00794																	2 He 4.00260
2	3 Li 6.941	4 Be 9.0122										5 B 10.811	6 C 12.011	7 N 14.0064	8 O 15.9994	9 F 18.99840	10 Ne 20.1797	
3	11 Na 22.989769	12 Mg 24.30467	III B (3)	IV B (4)	V B (5)	V I B (6)	V II B (7)	V III B (8) (9) (10)			IB (11)	II B (12)	13 Al 26.981538	14 Si 28.0855	15 P 30.97376	16 S 32.065	17 Cl 35.4527	18 Ar 39.948
4	19 K 39.0983	20 Ca 40.078	21 Sc 44.9559	22 Ti 47.88	23 V 50.9415	24 Cr 51.9961	25 Mn 54.9380	26 Fe 55.847	27 Co 58.9332	28 Ni 58.69	29 Cu 63.546	30 Zn 65.39	31 Ga 69.723	32 Ge 72.61	33 As 74.92159	34 Se 78.96	35 Br 79.904	36 Kr 83.80
5	37 Rb 85.4678	38 Sr 87.62	39 Y 88.9058	40 Zr 91.224	41 Nb 92.90638	42 Mo 95.94	43 Tc 98.9062	44 Ru 101.07	45 Rh 102.90550	46 Pd 106.42	47 Ag 107.8682	48 Cd 112.411	49 In 114.82	50 Sn 118.710	51 Sb 121.75	52 Te 127.60	53 I 126.90447	54 Xe 131.29
6	55 Cs 132.90543	56 Ba 137.327	57 *La 138.9055	72 Hf 178.49	73 Ta 180.9479	74 W 183.85	75 Re 186.207	76 Os 190.2	77 Ir 192.22	78 Pt 195.08	79 Au 196.96654	80 Hg 200.59	81 Tl 204.3833	82 Pb 207.2	83 Bi 208.98037	84 Po 209	85 At 209	86 Rn 222.0175
7	87 Fr 223.0187	88 Ra 226.0254	89 †Ac 227.0278	104 Rf 261.10	105 Ha 262.104	106 Sg 263.108	107 Ns 262.10	108 Hs (265)	109 Mt (266)	110 Uun (269)	111 Uuu (272)	112 Uub (277)	Halogens Noble or Inert Gases					

Alkali Metals Alkaline Earth Metals

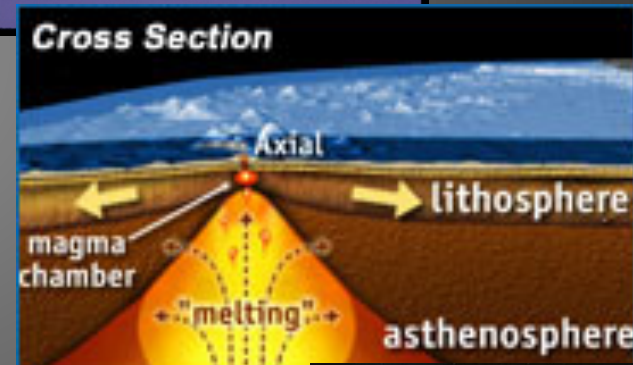
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Three Eras of Earth's Mineral Evolution

1. The Era of Planetary Accretion



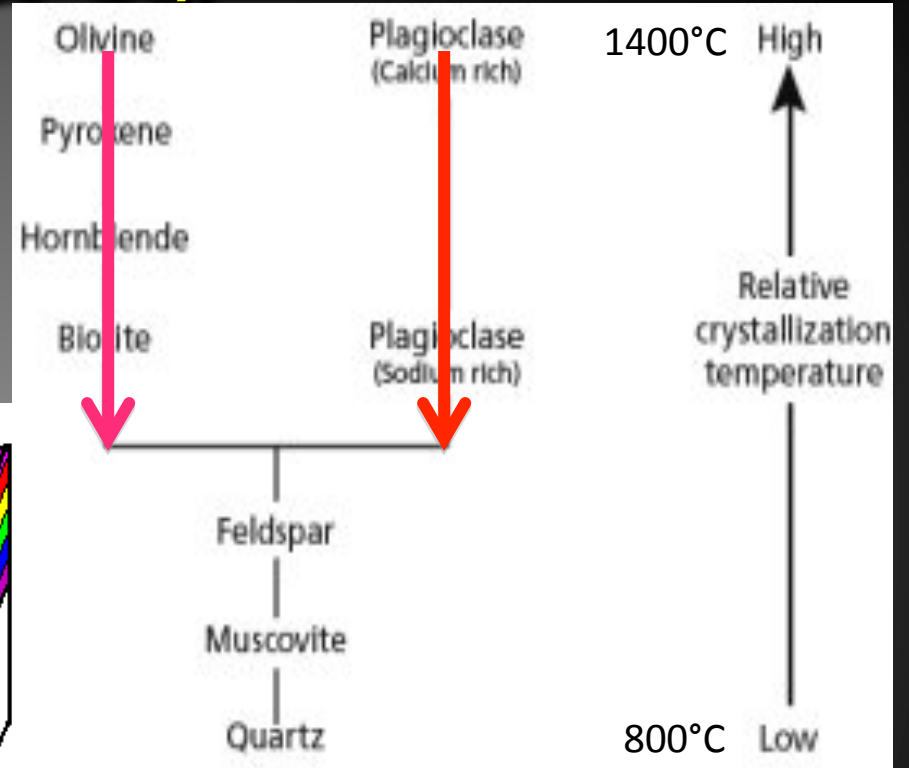
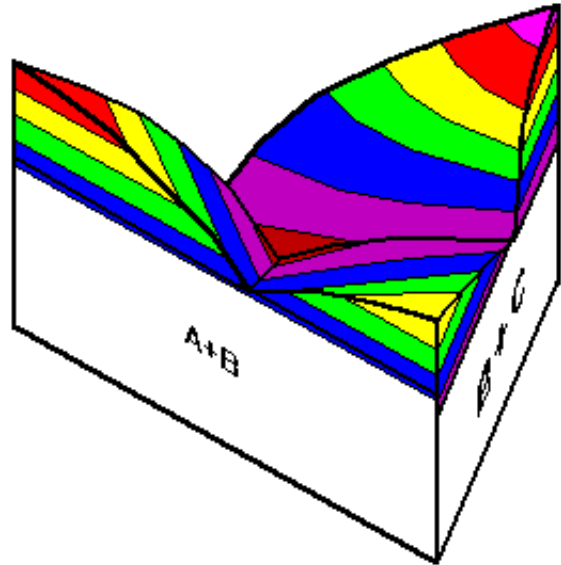
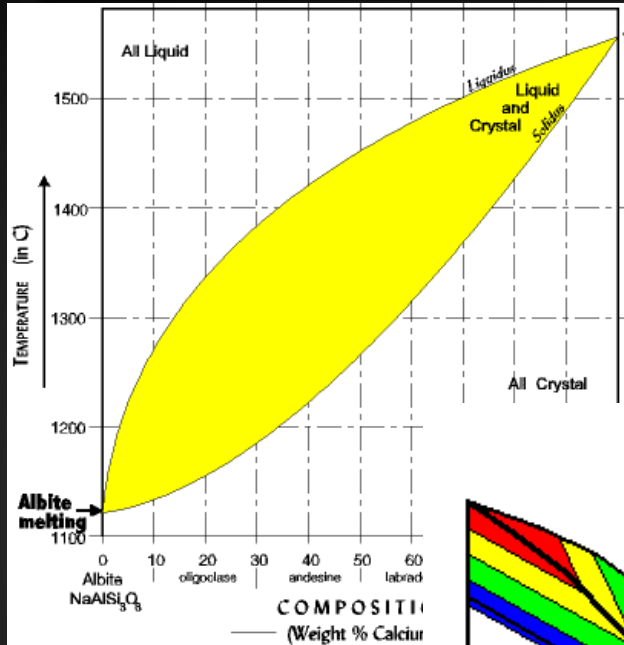
2. The Era of Crust and Mantle Reworking



3. The Era of Bio-Mediated Mineralogy



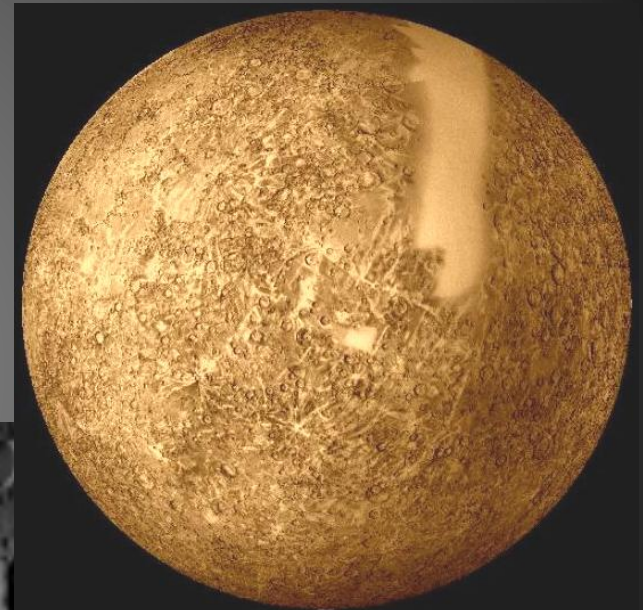
Stage 3: Initiation of Igneous Rock Evolution (4.55-4.0 Ga)



Partial melting, fractional crystallization and magma immiscibility

Stage 3: Initiation of Igneous Rock Evolution Volatile-poor Body

~350 mineral species?



Is this the end point of the Moon and Mercury?

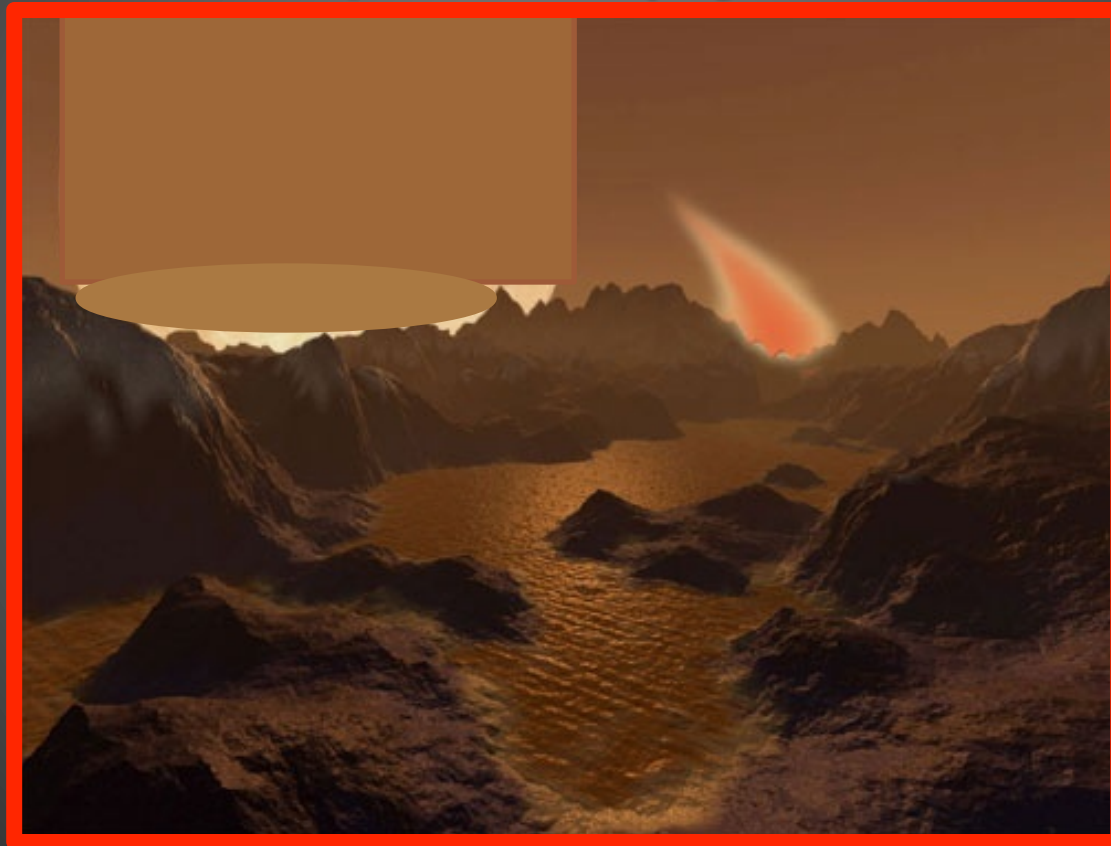
Stage 3: Initiation of Igneous Rock Evolution on a Volatile-rich Body (4.55-4.0 Ga)



Volcanism, outgasing and surface hydration.

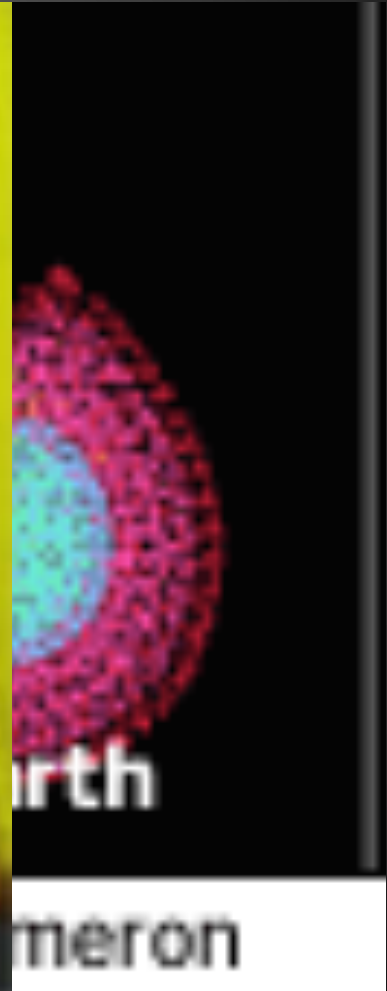
Stage 3: Initiation of Igneous Rock Evolution Volatile-rich Body

>500 mineral species (hydroxides, clays)



Volcanism, outgasing, surface hydration, evaporites, ices.

The Formation of the Moon



Stage 3: Initiation of Igneous Rock Evolution Volatile-rich Body

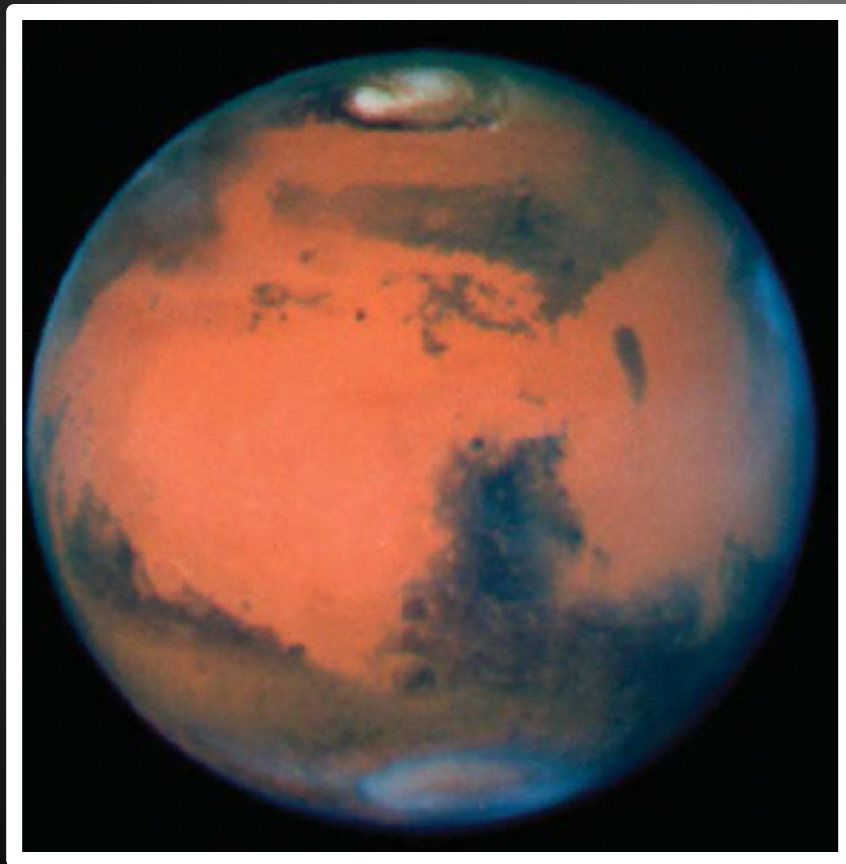
>500 mineral species (hydroxides, clays)



Volcanism, outgasing, surface hydration, evaporites, ices.

Stage 3: Initiation of Igneous Rock Evolution Volatile-rich Body

Is this the end point for Mars?



Volcanism, outgasing, surface hydration, evaporites, ices.

Stage 4: Granitoid Formation (>3.5 Ga)

>1000 mineral species (pegmatites)



Partial melting of basalt and/or sediments.

Stage 4 Minerals

Atomic number

Atomic mass

metal metalloid nonmetal

	IA (1)	IIA (2)	VIII B (8) (9) (10) (11) (12)										IIIA (13)	IVA (14)	VA (15)	VIA (16)	VIIA (17)	VIIIA (18)
1	1 H 1.00794																	2 He 4.00260
2	3 Li 6.941	4 Be 9.01218											5 B 10.811	6 C 12.011	7 N 14.0064	8 O 15.9994	9 F 18.99840	10 Ne 20.1797
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4 The periodic table of

Stage 4: Granitoid Formation (>3.5 Ga)

>1000 mineral species (pegmatites)



Tourmaline



Spodumene



Beryl



Pollucite

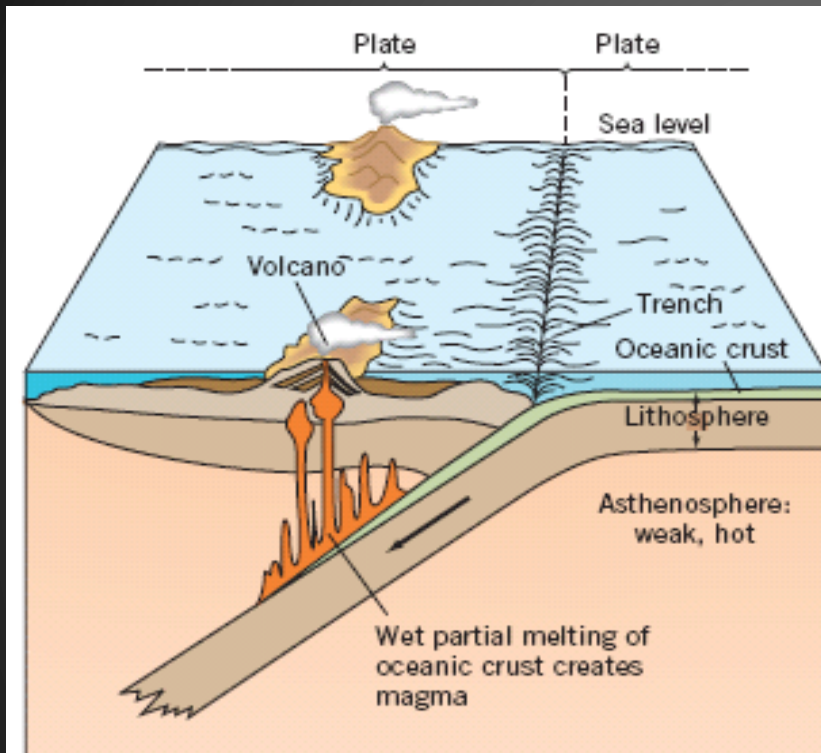


Tantalite

Complex pegmatites require multiple cycles of “eutectic” melting and fluid concentration.

Must they be younger than 3.5 Ga?

Stage 5: Plate tectonics and large-scale hydrothermal reworking of the crust (>3 Ga)



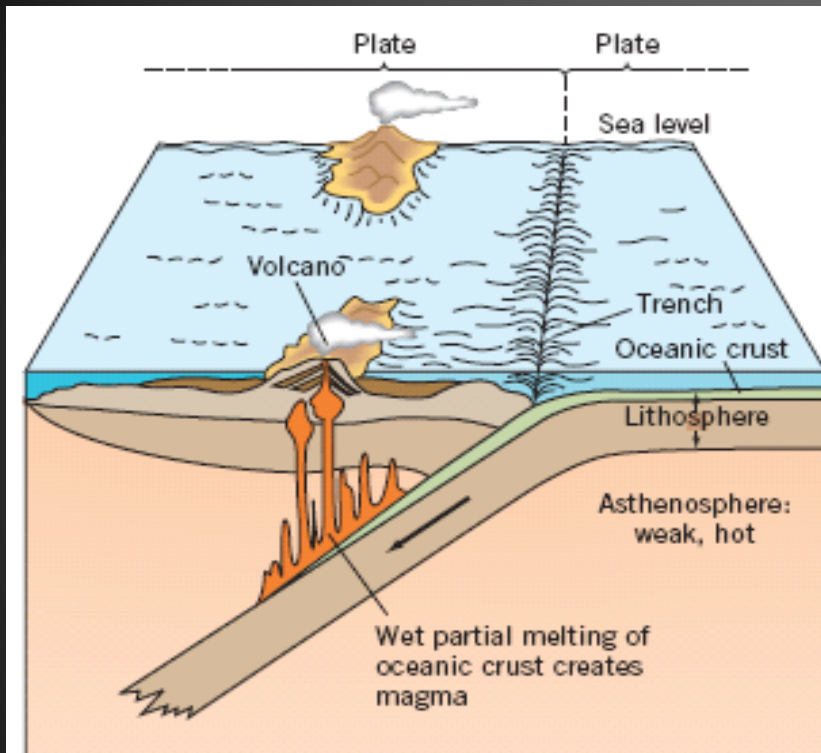
$\sim 10^8 \text{ km}^3$ of reworking



Mayon Volcano, Philippines

New modes of volcanism

Stage 5: Plate tectonics and large-scale hydrothermal reworking of the crust (>3 Ga)



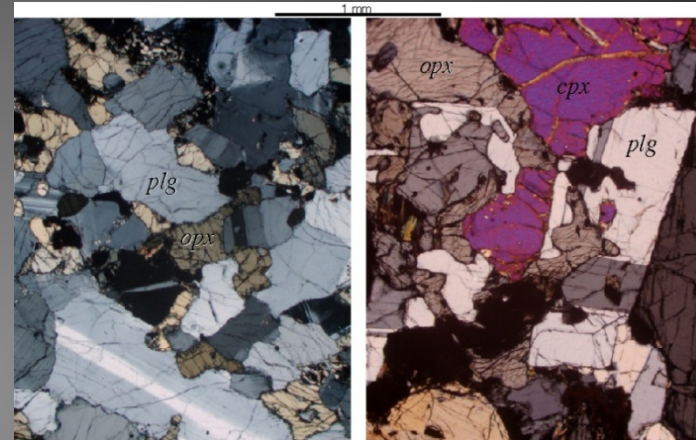
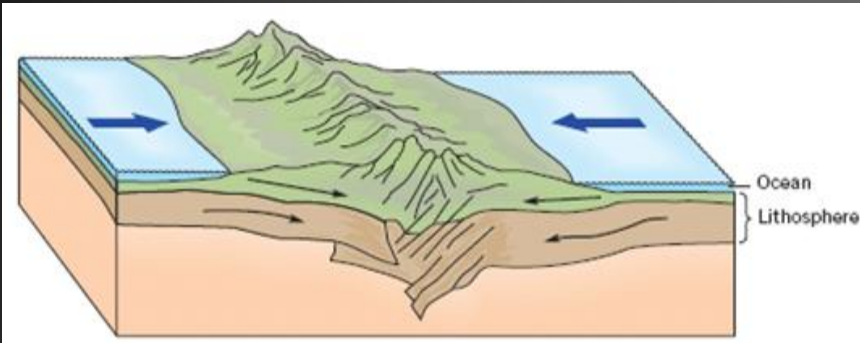
Rio Tinto. Spain

New modes of volcanism

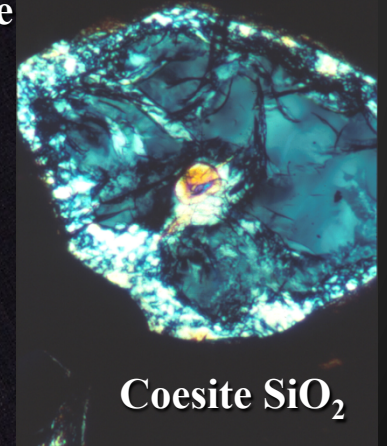
Massive base metal deposits (sulfides, sulfosalts)

Stage 5: Plate tectonics and large-scale hydrothermal reworking of the crust (>3 Ga)

1,500 mineral species



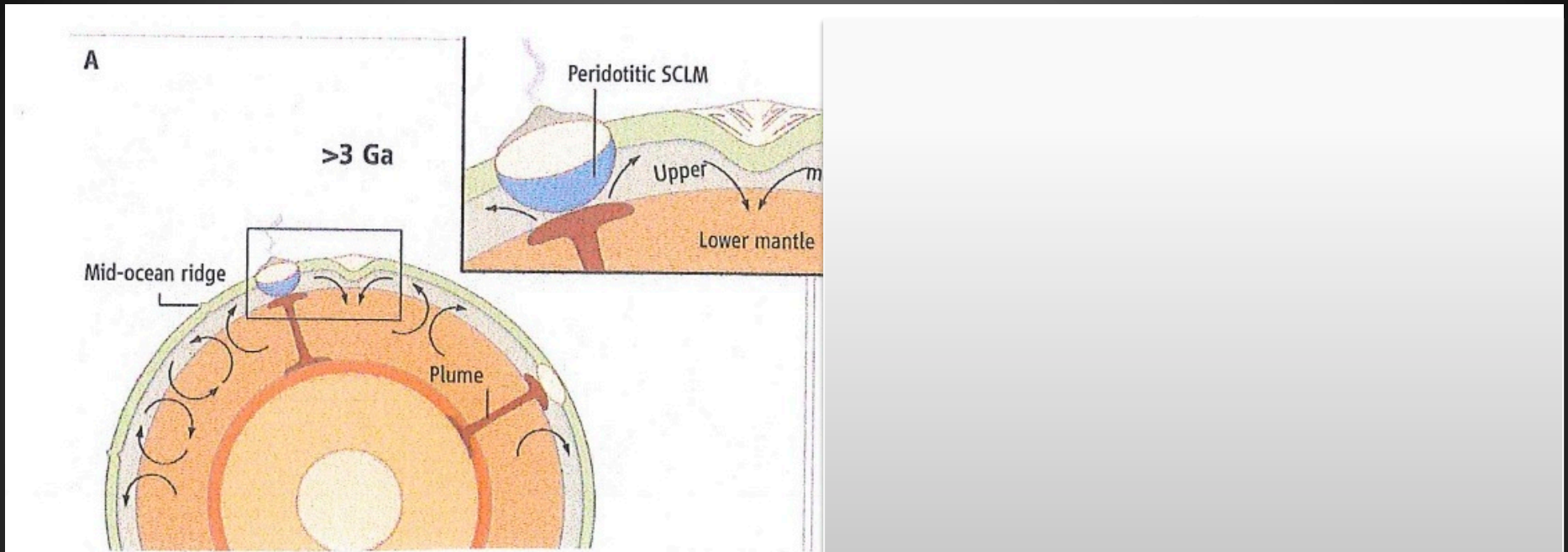
Glaucophane, Lawsonite, Jadeite



Coesite SiO₂

High-pressure metamorphic suites
(blueschists; granulites; UHP phases)

Stage 5: Plate tectonics and large-scale hydrothermal reworking of the crust (>3 Ga)

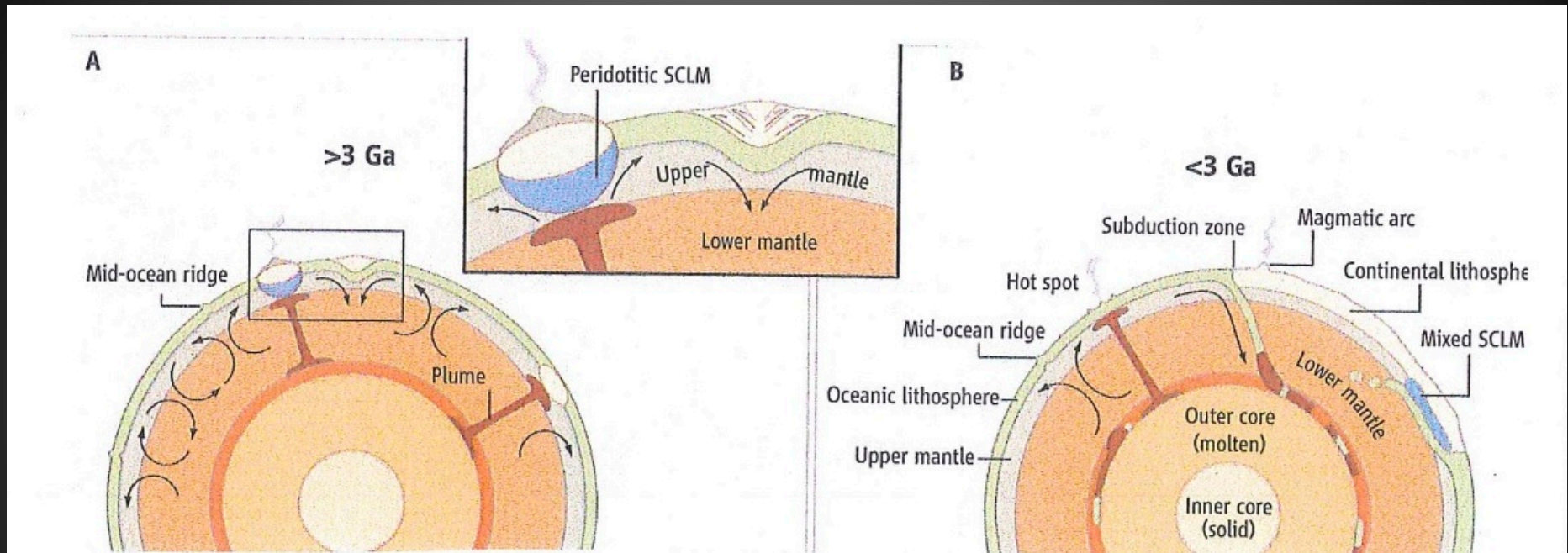


Van Kranendonk (2011)

> 3 Ga

When did subduction begin?

Stage 5: Plate tectonics and large-scale hydrothermal reworking of the crust (>3 Ga)



Van Kranendonk (2011)

> 3 Ga

< 3 Ga

When did subduction begin?

Stages 3-5: Era of crust-mantle processing (igneous evolution; plate tectonics)

New geologic processes, especially fluid-rock interactions associated with igneous activity and plate tectonics, led to a greater diversity of geochemical environments and thus new mineral species.

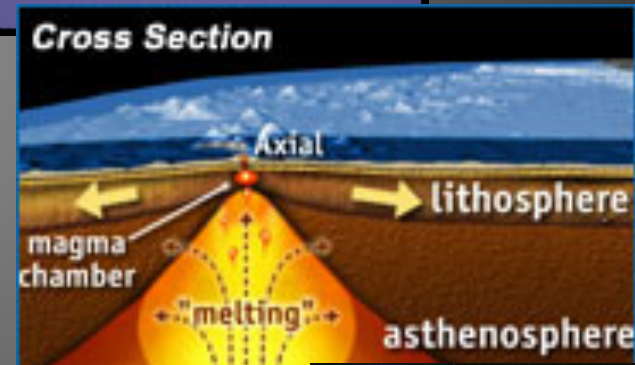
~1500 mineral species

Three Eras of Earth's Mineral Evolution

1. The Era of Planetary Accretion



2. The Era of Crust and Mantle Reworking



3. The Era of Bio-Mediated Mineralogy



The origin of life may require some minimal degree of mineral evolution.

Sulfides



Clays



Borates



Conversely, does further mineral evolution depend on life?

Hence the co-evolution of the geo- and biospheres.

Stage 6: Anoxic Archean biosphere (3.9-2.5 Ga)

~1,500 mineral species (BIFs, carbonates,



Photo credit: D. Papineau



Photo credit: F. Corsetti, USC

Stage 6: Anoxic Archean biosphere (3.9-2.5 Ga)

~1,500 mineral species (BIFs, carbonates, sulfates, evaporites, skarns)



Death Valley evaporites
(courtesy Smith College)



Grossular



Diopside



Idocrase

Stage 7: Paleoproterozoic Oxidation (2.5-1.85 Ga)

>4500 mineral species, including perhaps
>2,800 new oxides/hydroxides



Rise of oxidative photosynthesis.

Stage 7: Paleoproterozoic Oxidation (2.5-1.85 Ga)

>4500 mineral species, including perhaps
>2,800 new oxides/hydroxides



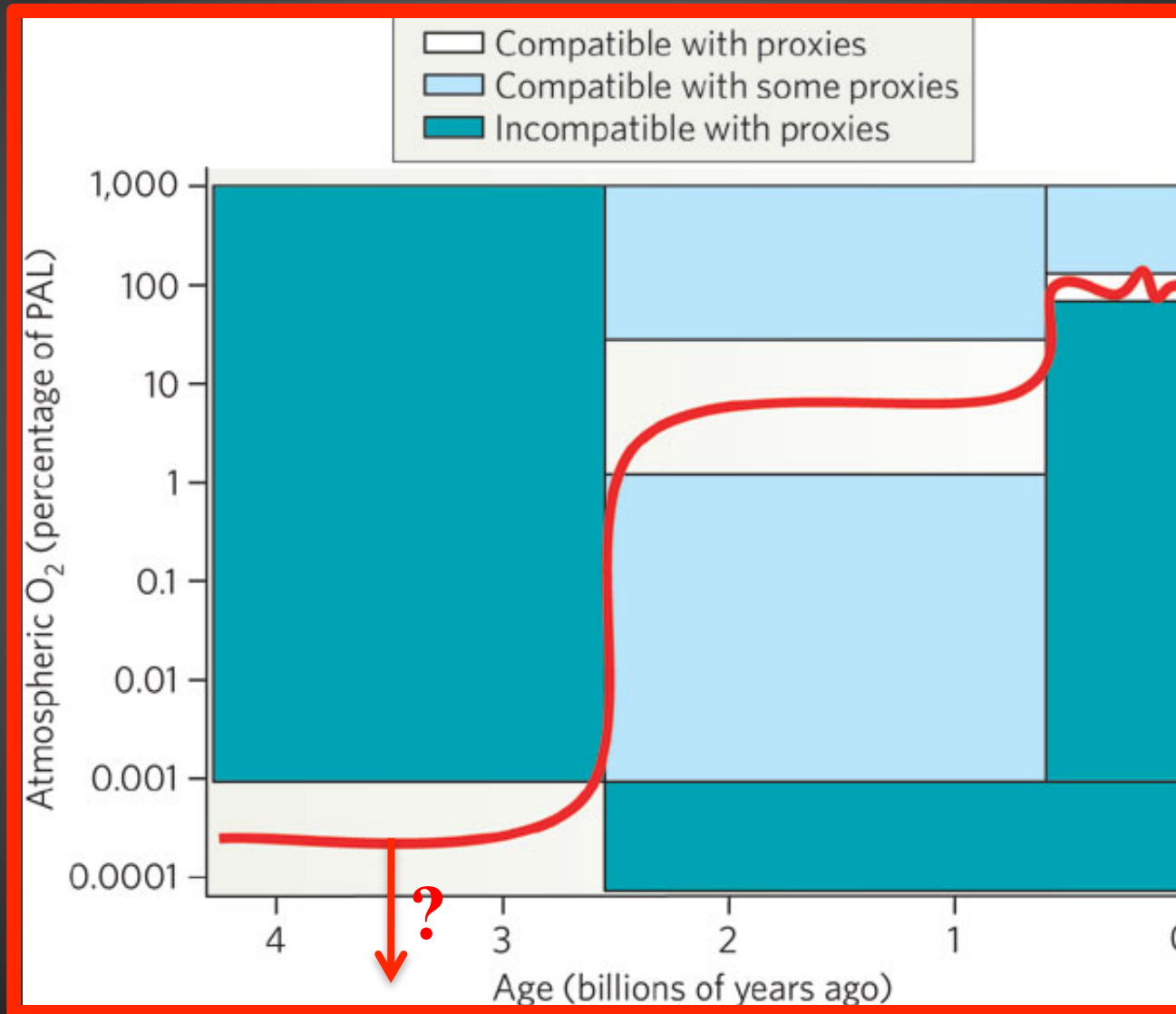
Rise of oxidative photosynthesis.

Hypothesis

Approximately 2/3rds of all known mineral species cannot form in an anoxic environment, and thus are the indirect consequence of biological activity.

Many lines of evidence point to an Archean atmosphere that was essentially lacking in oxygen.

What was the oxygen level in the Archean Eon?



Kump (2008) *Nature* 451, 277-278.

What was the oxygen level in the Archean Eon (> 2.5 billion years)?

Published estimates of log O₂ Pressure

Ohmoto (numerous refs)	> -2
Farquhar et al. (2000)	< -5
Frimmel (2005)	< -5
Kump (2008)	< -5
C-W-K-H Model (1968+)	~ -13
Sverjensky et al. (2008, 2010)	~ -70

Key constraints on Archean surface oxygen levels.

Detrital uraninite and pyrite

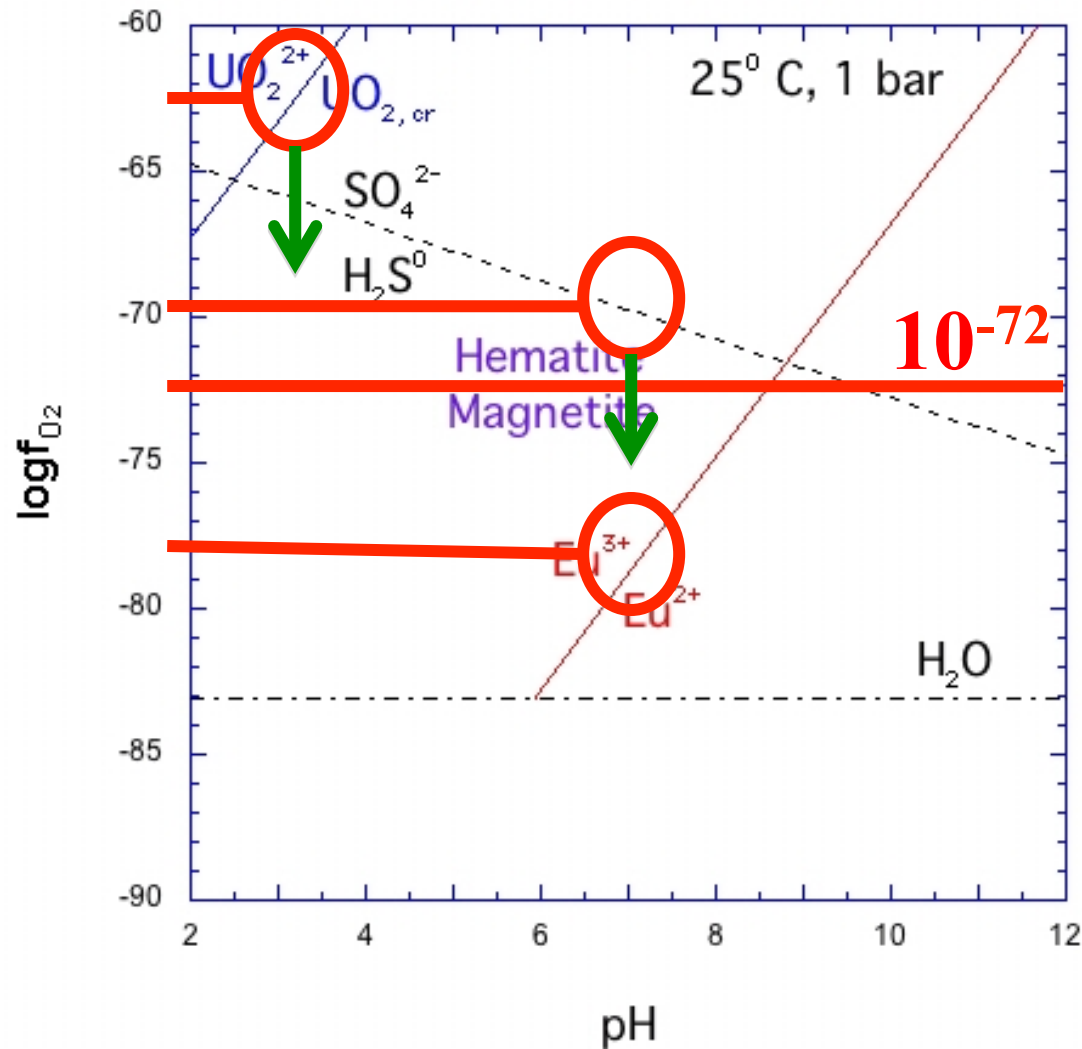
Paleosols lacking iron oxides

[Surface waters with aqueous Fe^{2+}]

[Surface waters with low SO_4^{2-}]

Eu^{2+} anomalies

What was the oxygen level in the Archean Eon?



Key constraints on Archean surface oxygen levels.

Detrital uraninite, pyrite and siderite

Paleosols lacking iron oxides

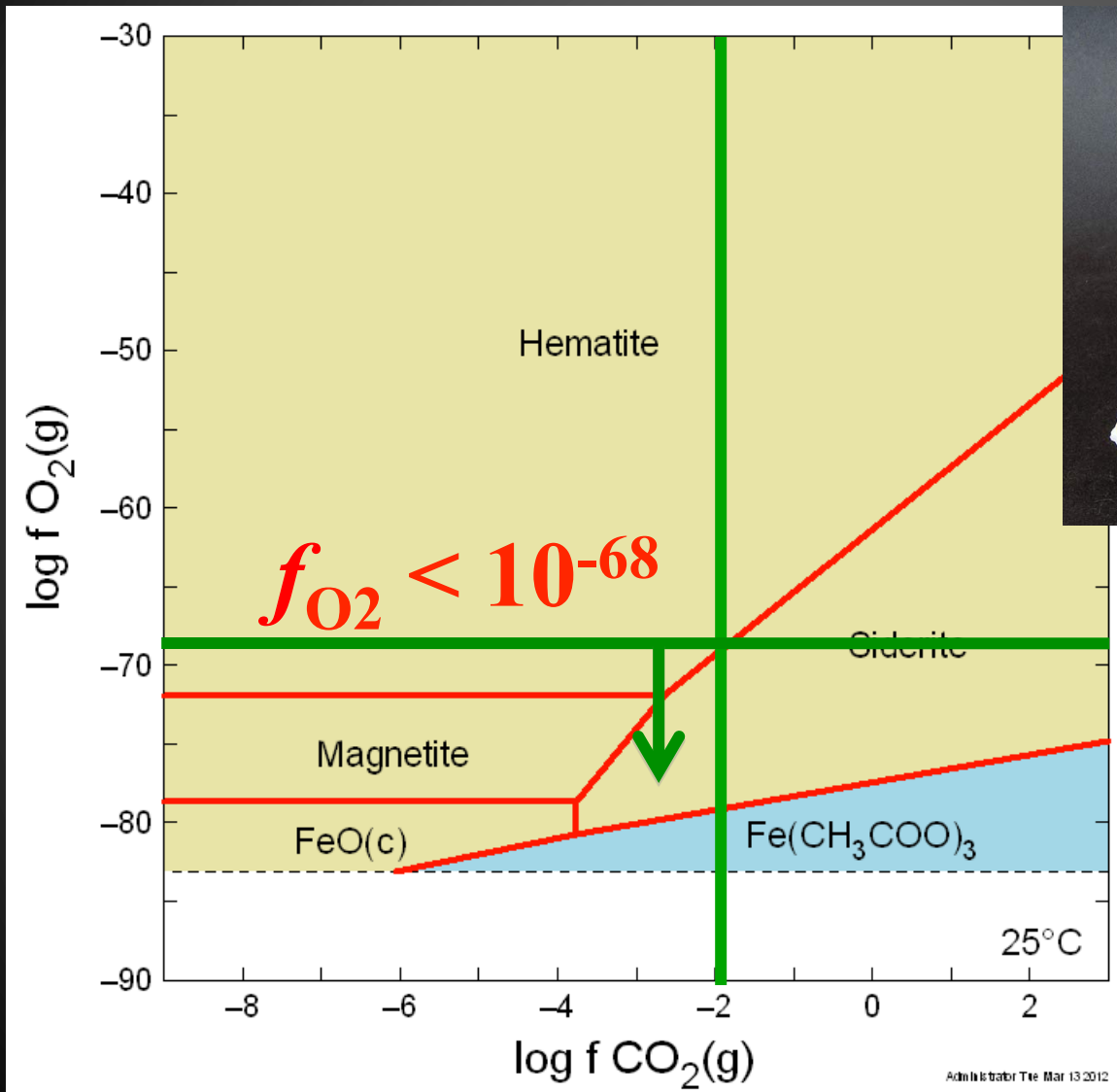
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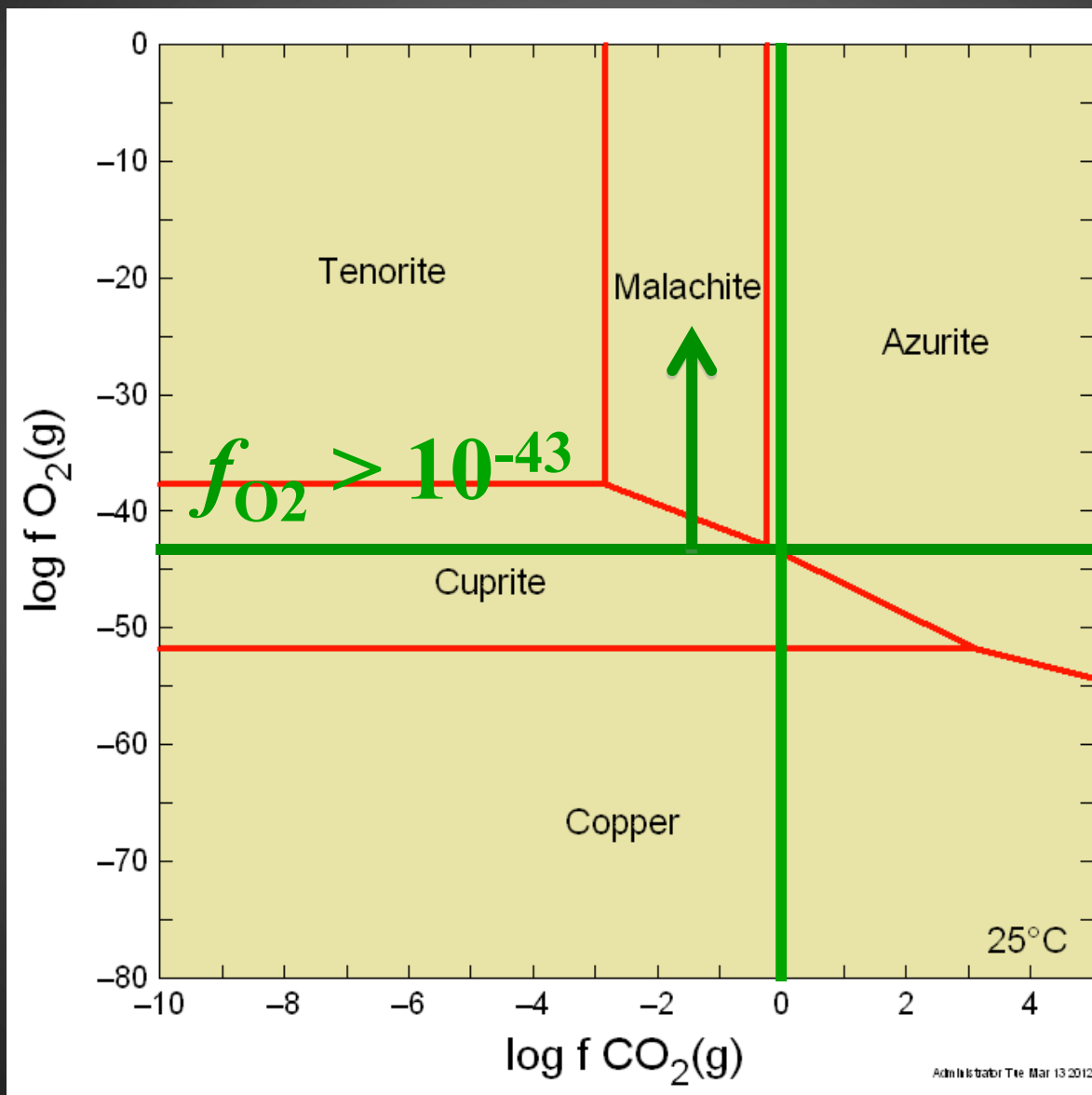
Precipitation of ferroan carbonates

What was the oxygen level in the Archean Eon?



Siderite
 $FeCO_3$

What was the oxygen level in the Archean Eon?



Azurite
&
Malachite



Stage 7: Paleoproterozoic Oxidation (2.5-1.85 Ga)

Cu^{2+} Copper minerals (256 of 321)



When did these minerals first appear?

Stage 7: Paleoproterozoic Oxidation (2.5-1.85 Ga)

What mineral species won't form?

202 of 220 U minerals

319 of 451 Mn minerals

47 of 56 Ni minerals

582 of 790 Fe minerals



Piemontite



CARNOTITE



Garnierite



Xanthoxenite

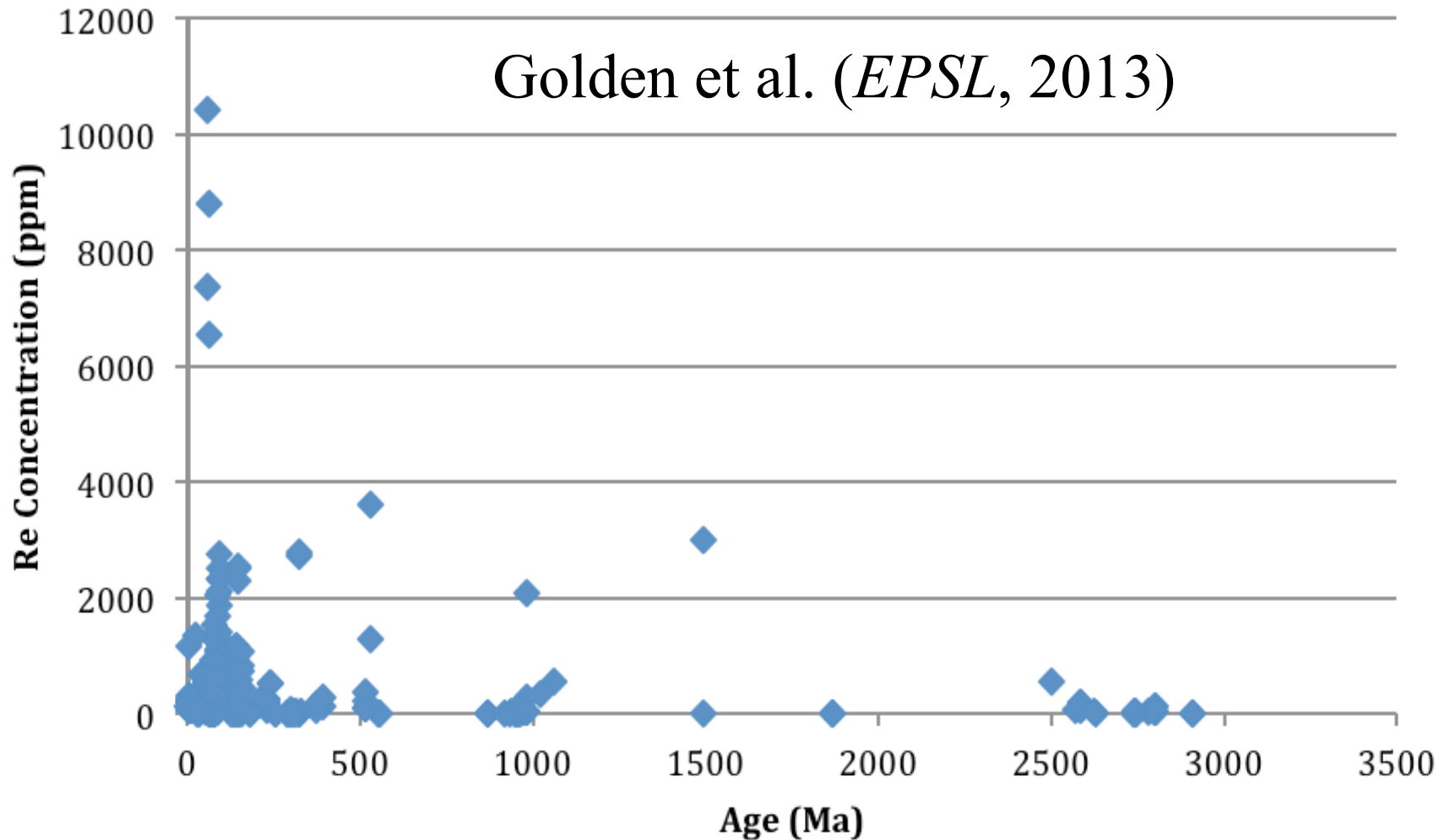
Stages 6-10: Co-evolution of the geosphere and biosphere

Changes in Earth's atmospheric composition at ~2.4 to 2.2 Ga represent the single most significant factor in our planet's mineralogical diversity.

>4500 mineral species

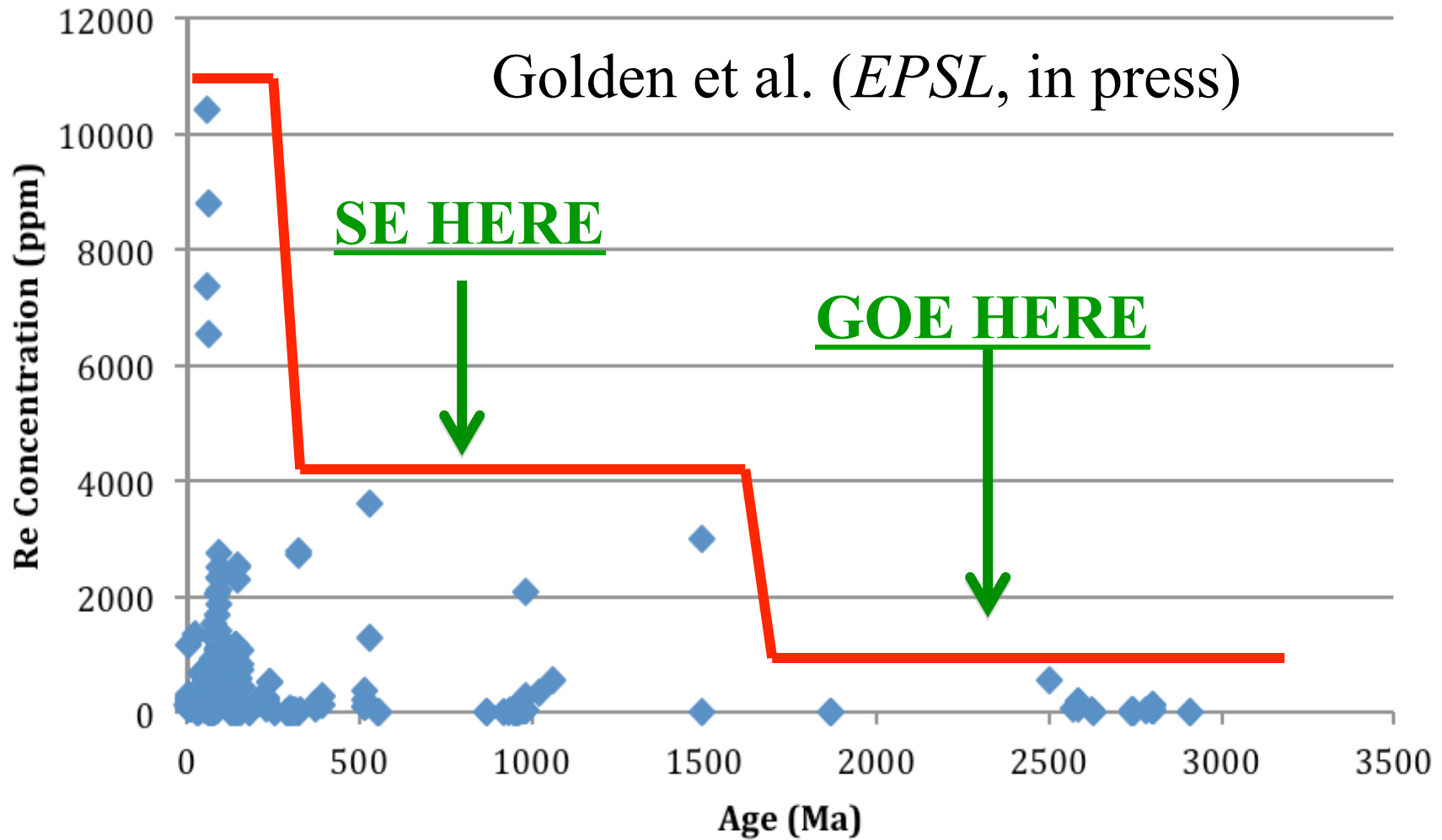
RESULTS: Molybdenite (MoS_2) through Time

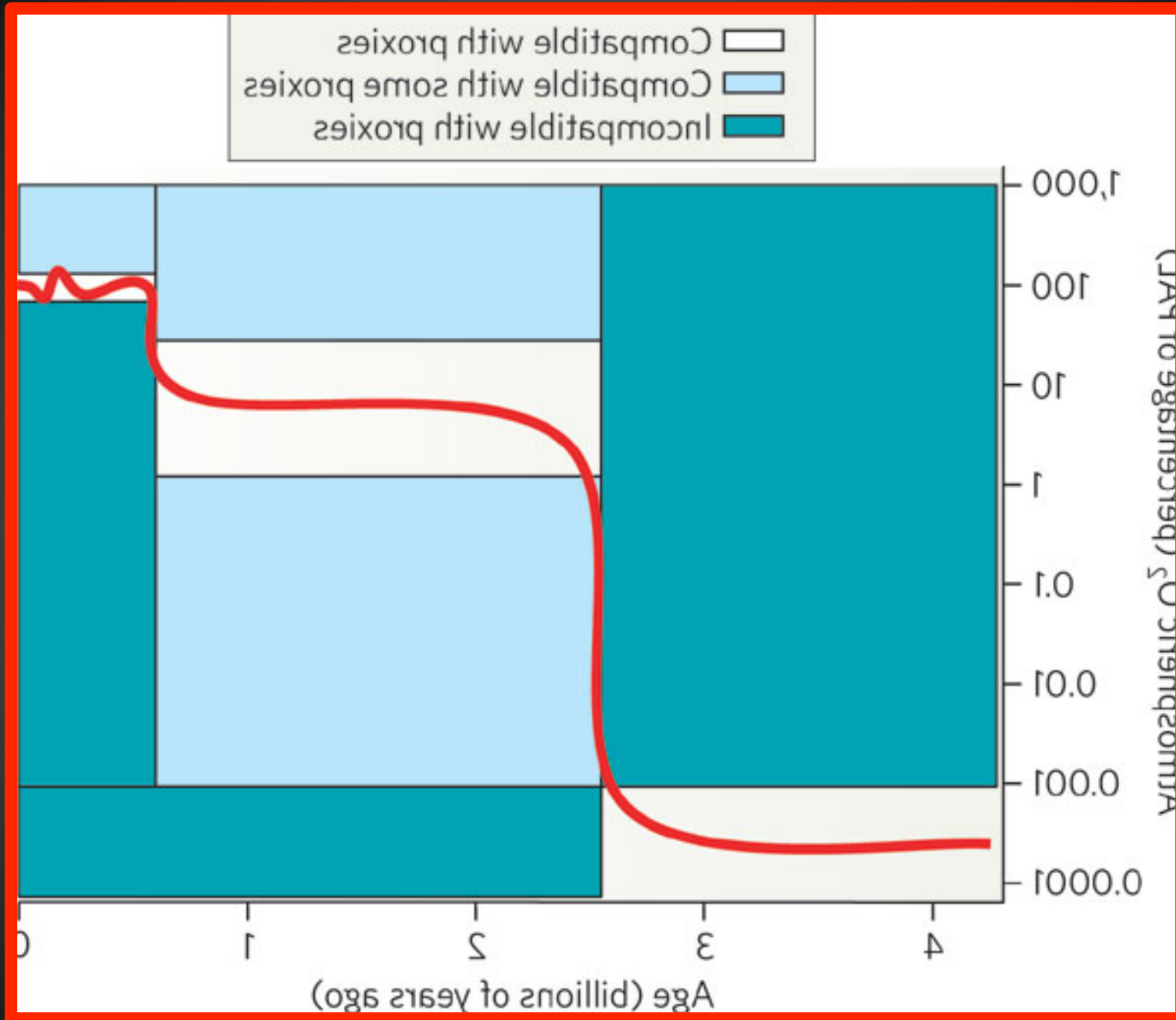
Re Concentration Vs. Time



RESULTS: Molybdenite (MoS_2) through Time

Re Concentration Vs. Time





Kump (2008) *Nature* 451, 277-278.

RESULTS: Molybdenite (MoS_2) through Time

Hypothesis: There was a protracted “Great Subsurface Oxidation Interval” that postdated the GOE by a billion years. This interval was the single most significant factor in Earth’s mineralogical diversification.

Stage 8: The “Intermediate Ocean” (1.85-0.85 Ga)

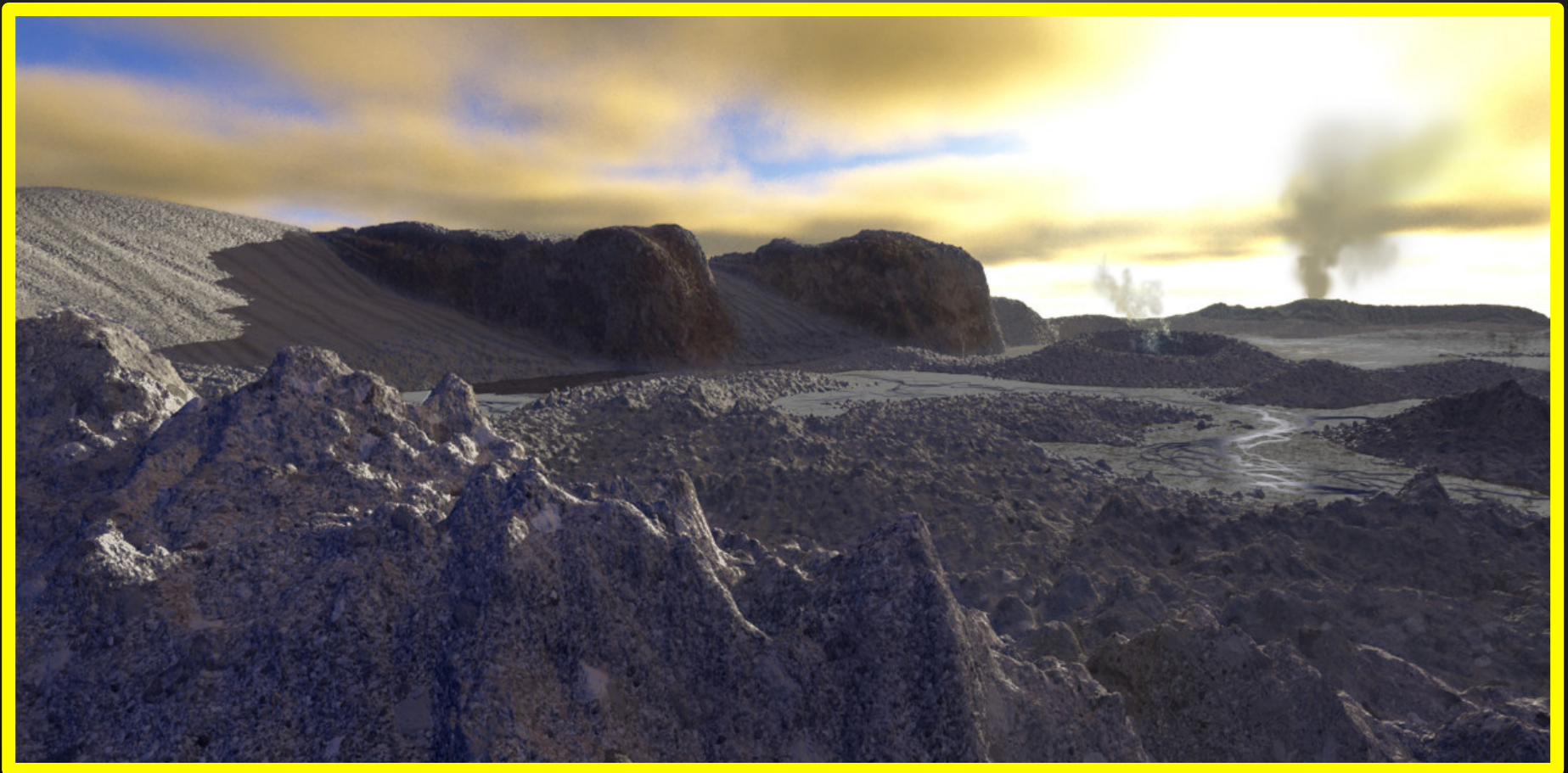
>4500 mineral species (few new species)



Oxidized surface ocean; deep-ocean anoxia.

Stage 9: Snowball Earth and Neoproterozoic Oxidation (0.85-0.542 Ga)

>4500 mineral species (few new species)



Glacial cycles triggered by albedo feedback.

Stage 10: Phanerozoic Biomineralization (<0.542 Ga)

>4,800 mineral species (Biominerals, clays)



Stage 10: Phanerozoic Biomineralization (<0.542 Ga)

>4,800 mineral species



Stage 10: Phanerozoic Biomineralization



Abelsonite— $\text{NiC}_{31}\text{H}_{32}\text{N}_4$



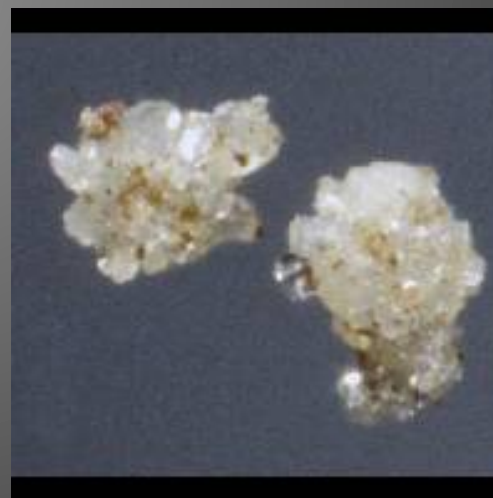
Ravatite— $\text{C}_{24}\text{H}_{48}$



Evankite— $\text{C}_{24}\text{H}_{48}$



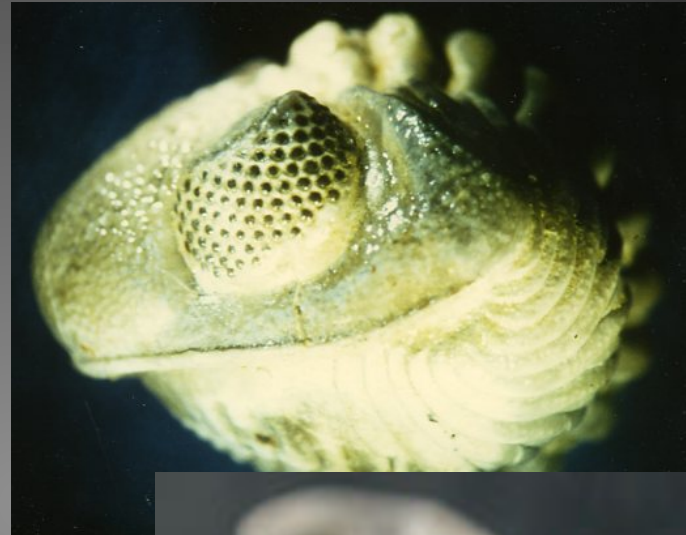
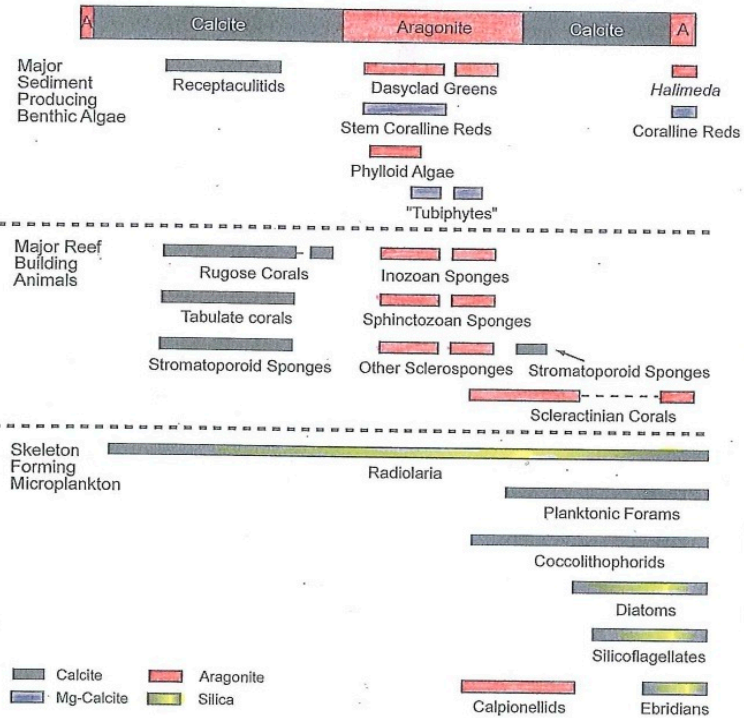
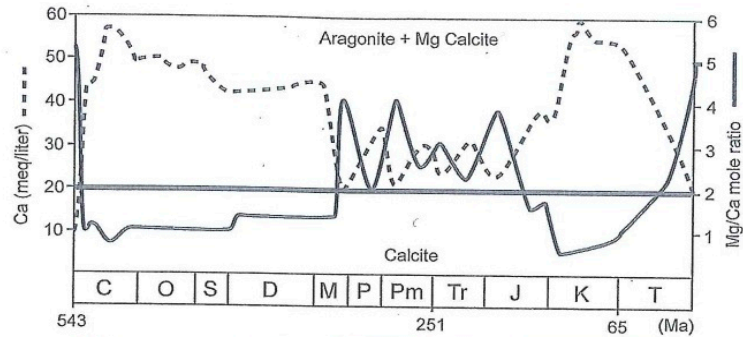
Dashkovaite— $\text{Mg}(\text{HCOO})_2 \cdot 2\text{H}_2\text{O}$



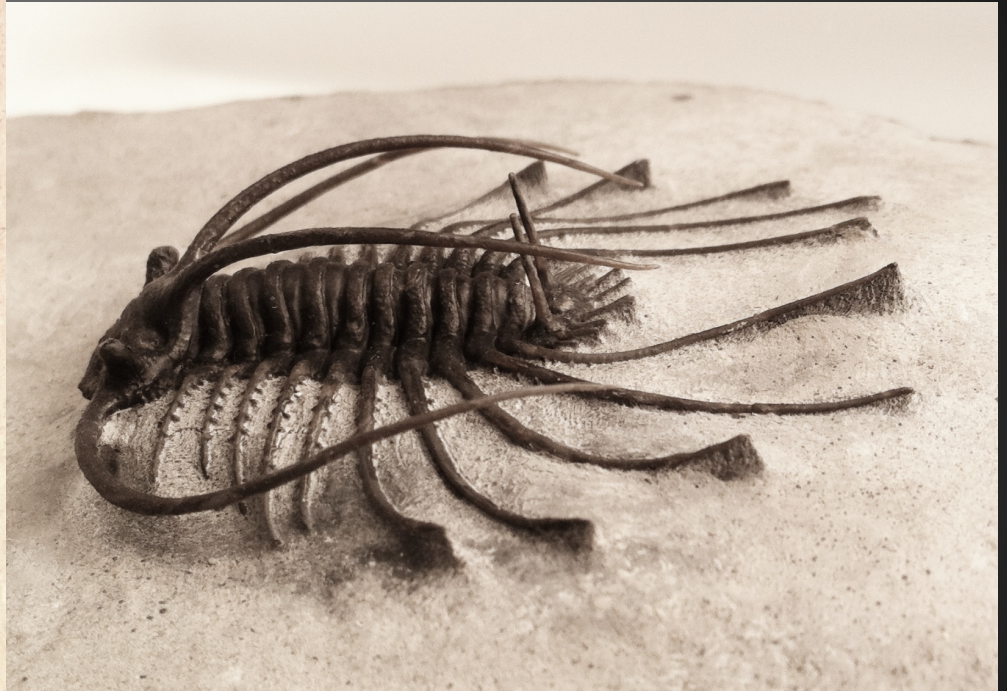
Oxammite— $(\text{NH}_4)(\text{C}_2\text{O}_4) \cdot \text{H}_2\text{O}$

> 50 Organic Mineral Species

Skeletal Biomineralization



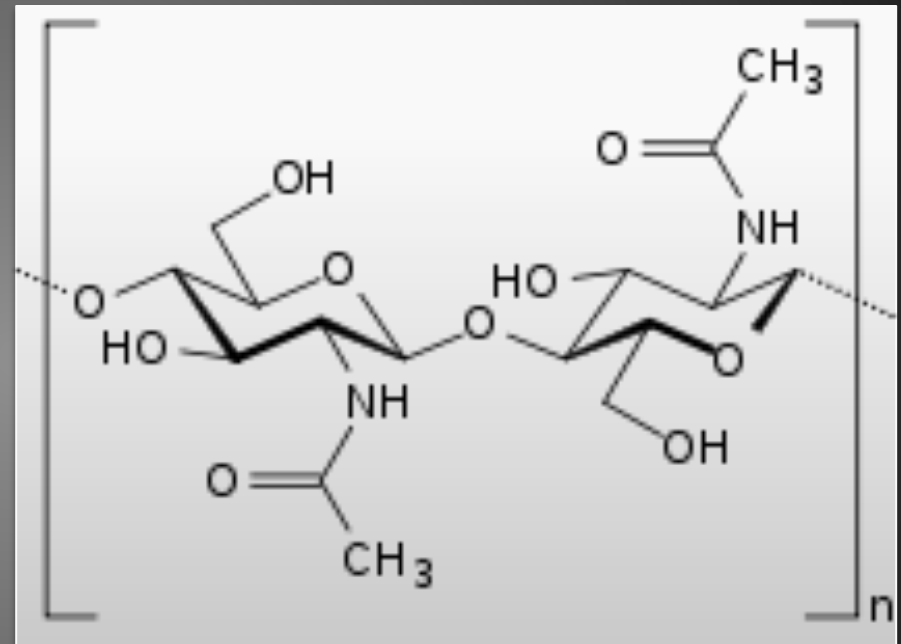
Stage 10: Phanerozoic Biomineralization



Apianurus nov. sp.

*Walcott-Rust Quarry,
Moscow, New York.*

Stage 10: Phanerozoic Biomineralization



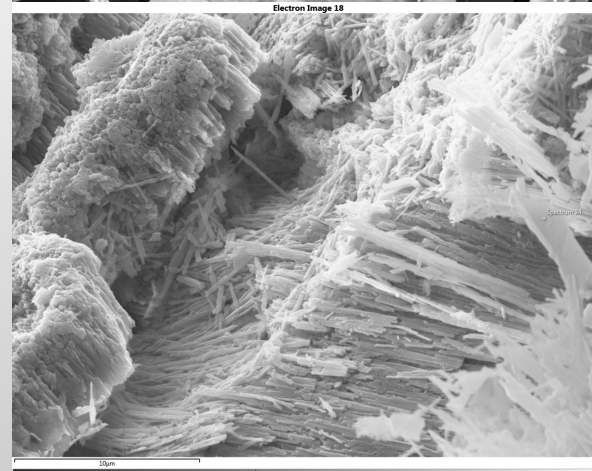
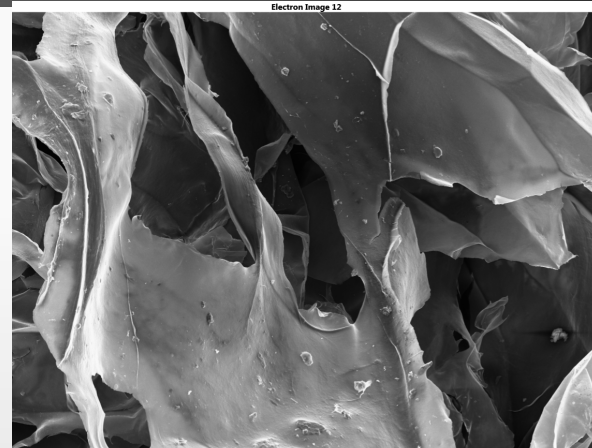
Apianurus nov. sp.

Walcott-Rust specimens preserve biomolecular fragments of chitin.

Stage 10: Phanerozoic Biomineralization



Ecphora



SEM images

Calvert Cliffs specimens preserve the original 15 million year old shell-binding proteins!

RECENT CONCLUSIONS

Previously unrecognized patterns in the distribution of minerals through Earth history reflect:

- The supercontinent cycle.
- The rise of the terrestrial biosphere.

The Supercontinent Cycle



PERMIAN
225 million years ago



TRIASSIC
200 million years ago



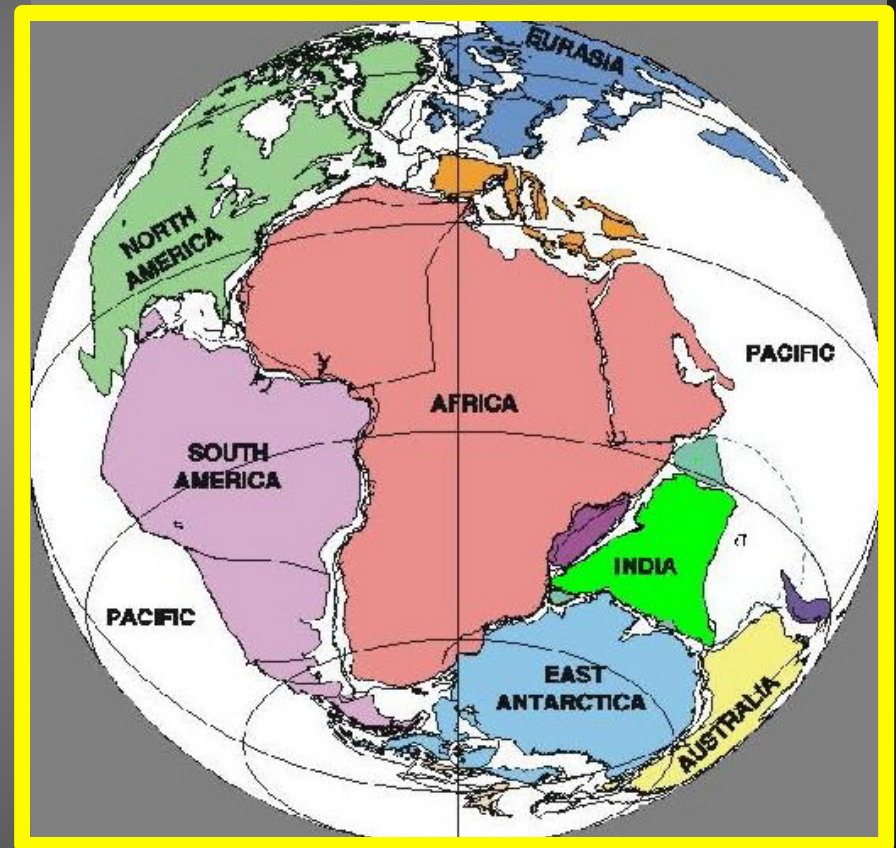
JURASSIC
135 million years ago



CRETACEOUS
65 million years ago



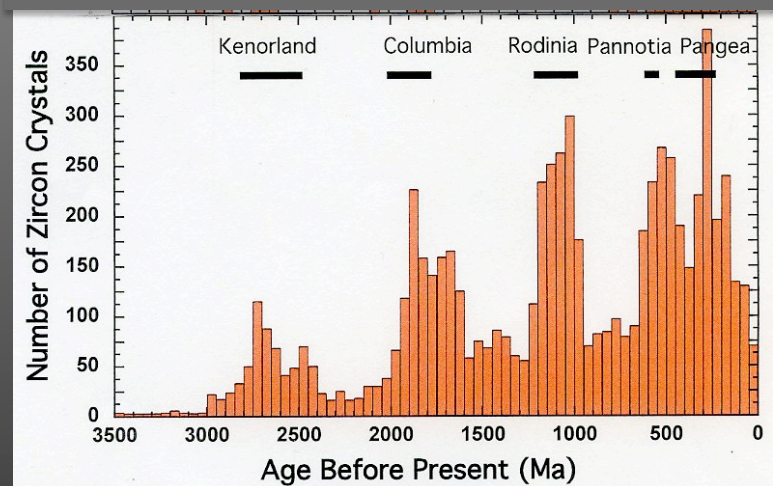
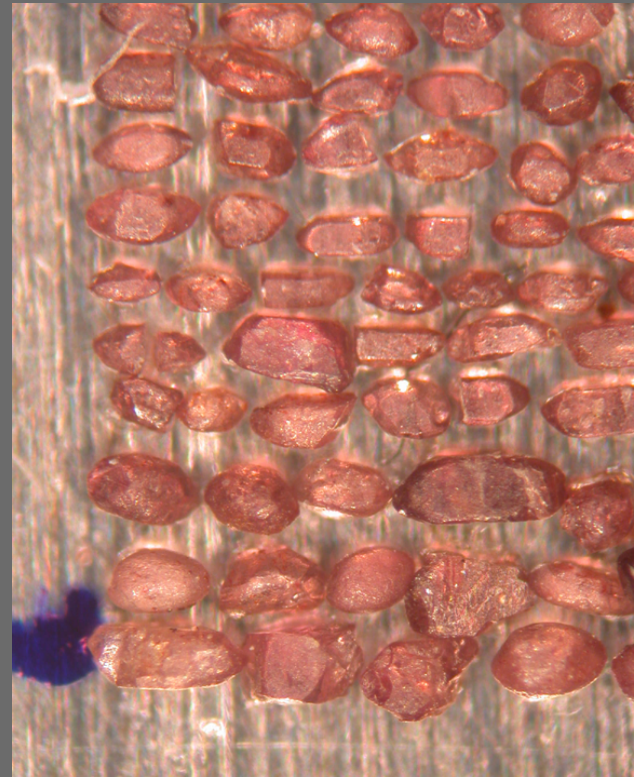
PRESENT DAY



RESULTS: The Supercontinent CYCLE

The distribution of zircon crystals through time correlates with the supercontinent cycle over the past 3 billion years.

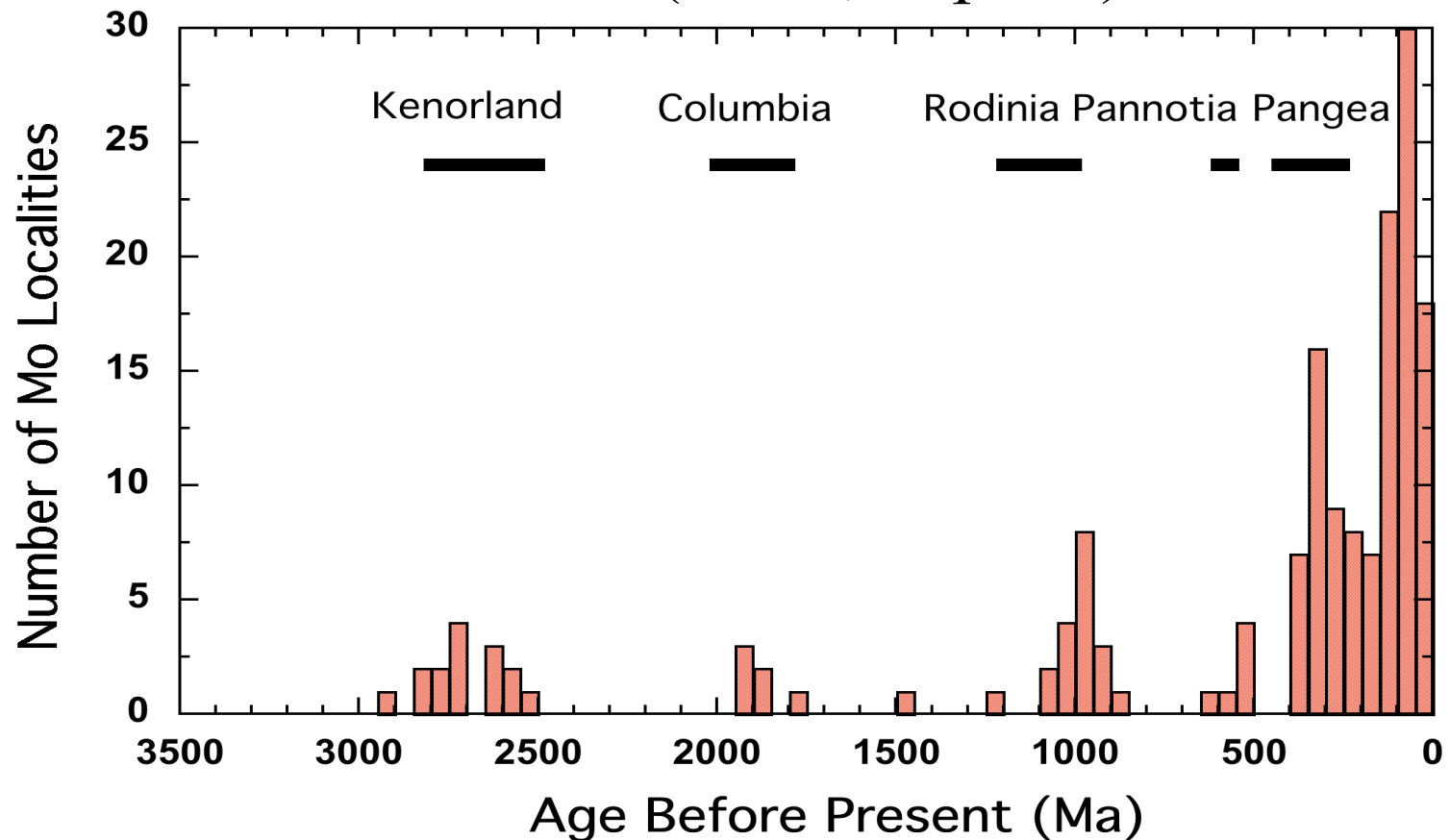
(Condie & Aster 2010;
Hawksworth et al. 2010)



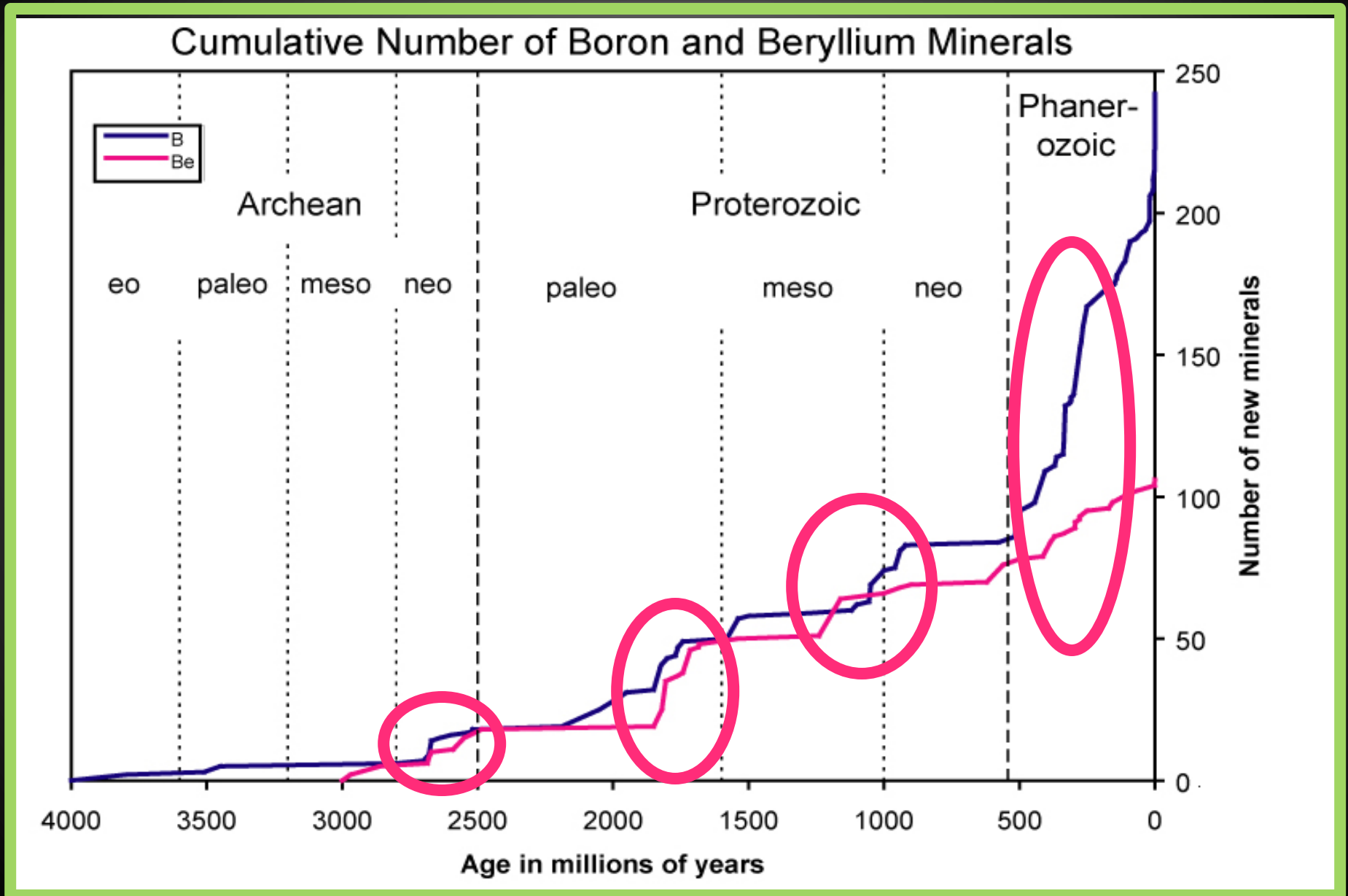
RESULTS: Mo Mineral Evolution

Temporal distribution of molybdenite (MoS_2)

Golden et al. (*EPSL*, in press)



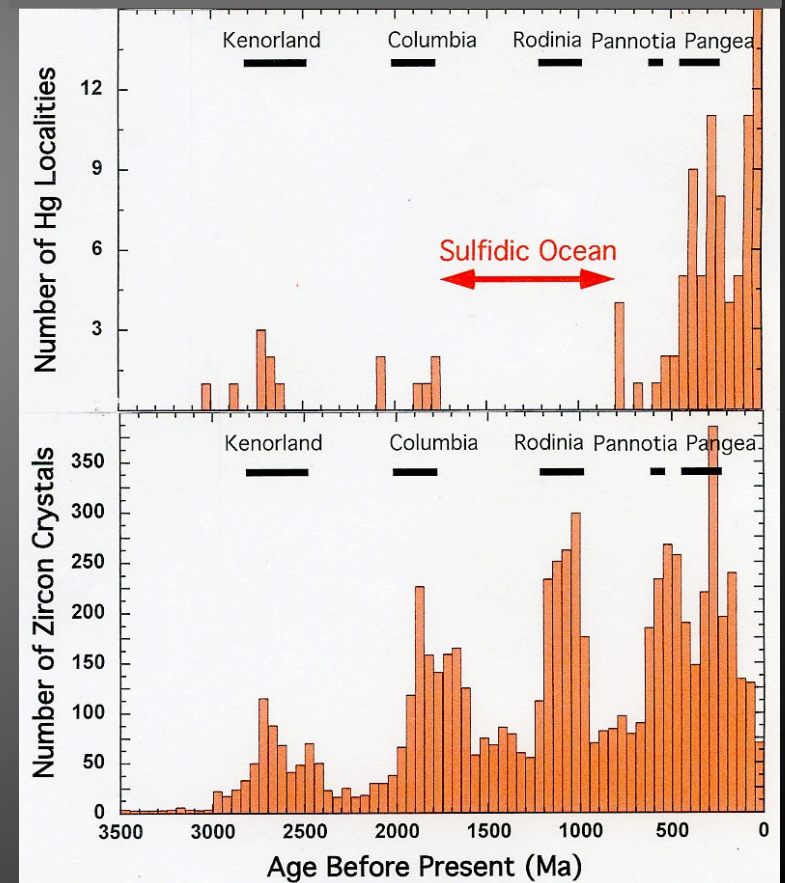
B and Be Minerals (Grew & Hazen 2009, 2010)



Hg Mineral Evolution

The distribution of mercury (Hg) minerals through time also correlates with the SC cycle over the past 3 billion years, but there's a gap during the "boring billion".

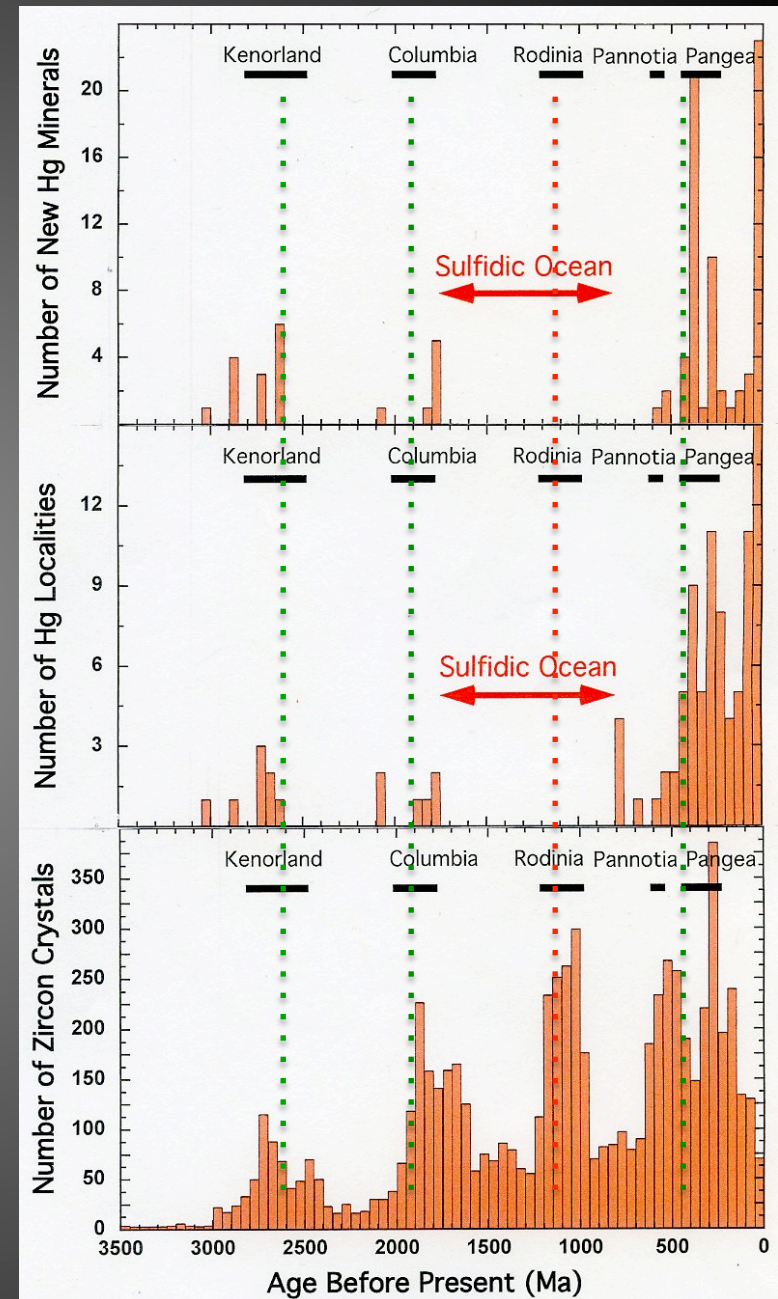
Hazen et al. (2012) *Amer. Mineral.*



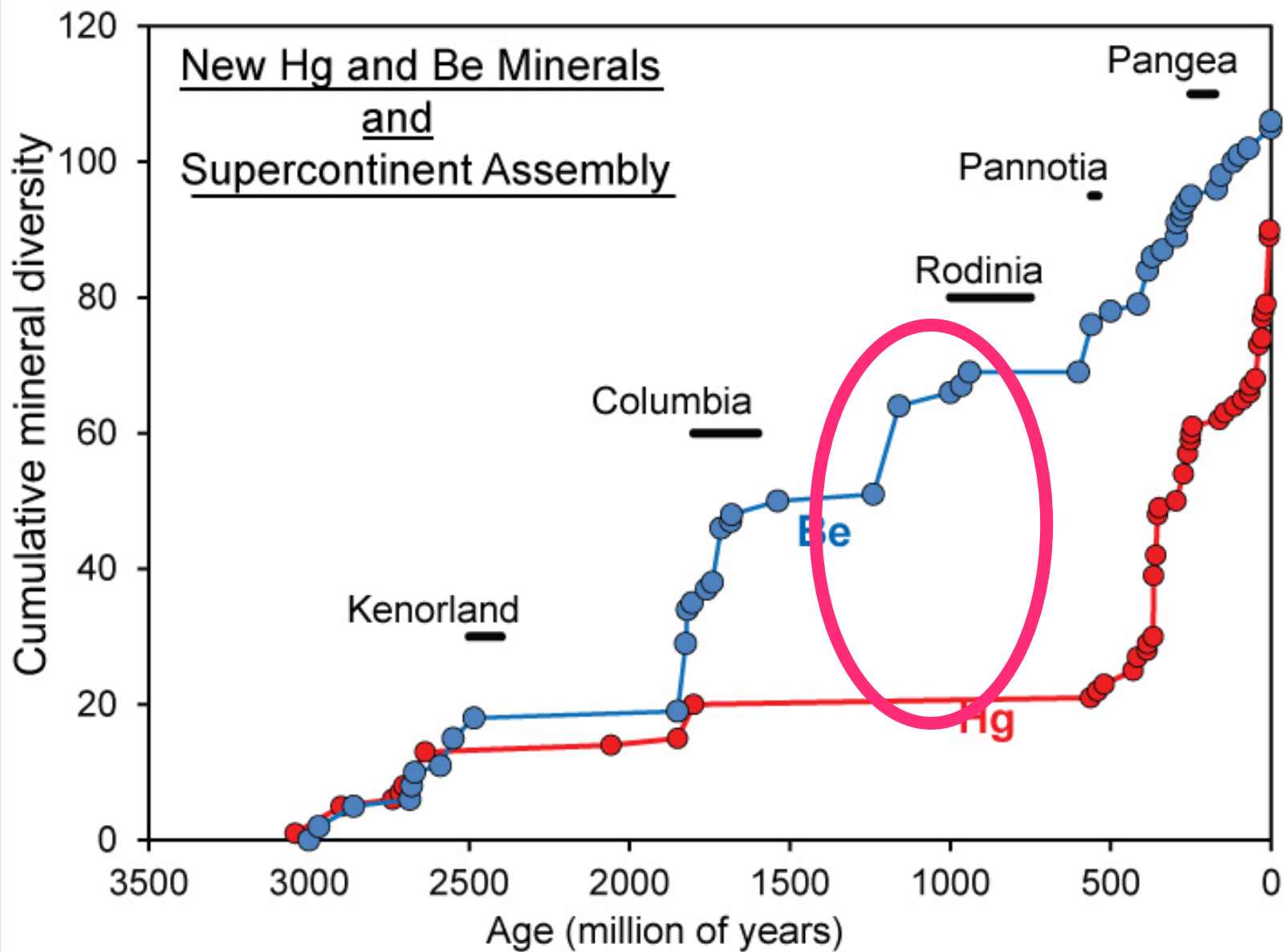
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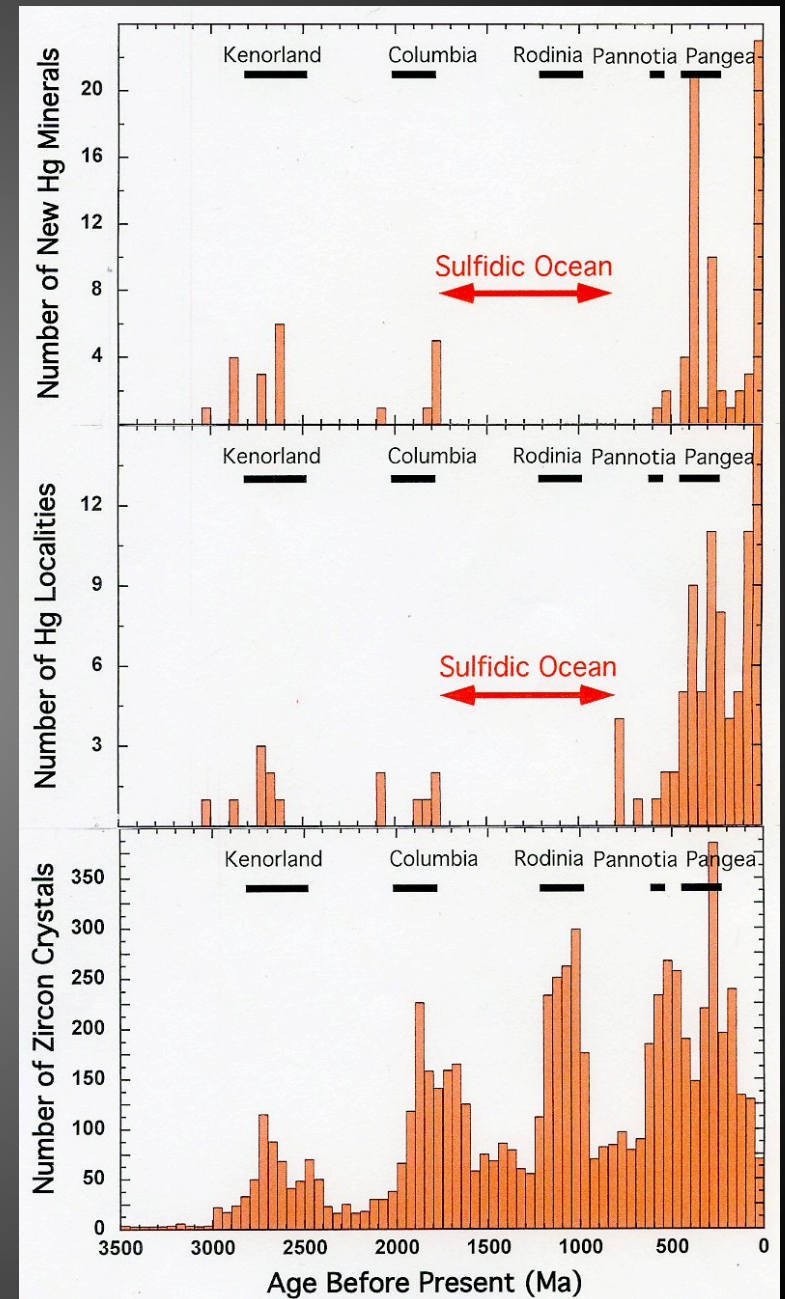
RESULTS: Hg & Be Mineral Evolution



Hg Mineral Evolution

1. Was Hg sequestered as insoluble cinnabar?
2. Did microbial methylation play a role?
3. Is Hg in Mesoproterozoic Pb-Zn deposits?

Hazen et al. (2012) *Amer. Mineral.*

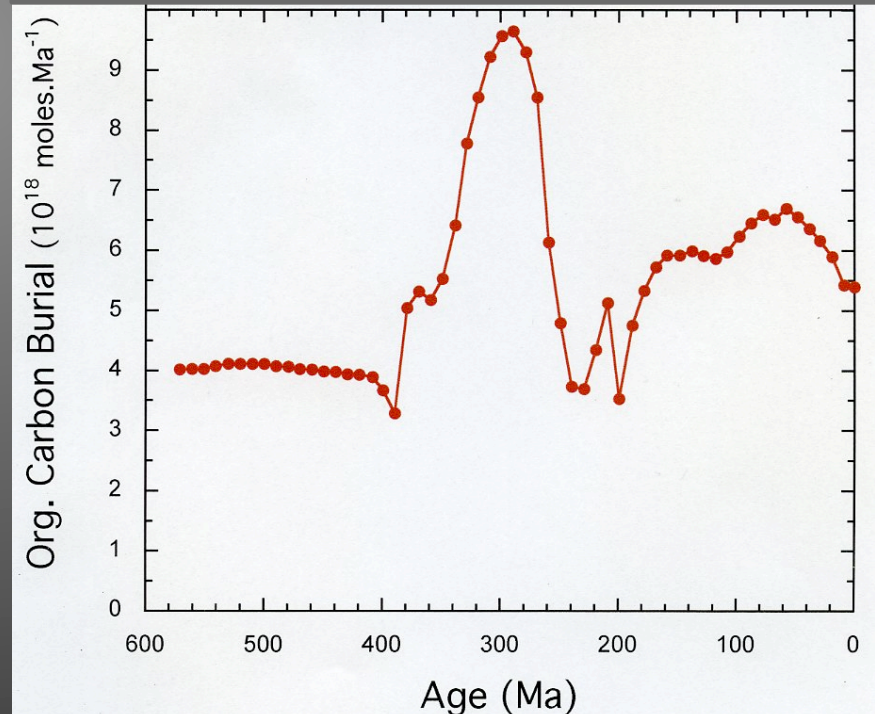


Phanerozoic Biomineralization and the Terrestrial Biosphere (<542 Ma)



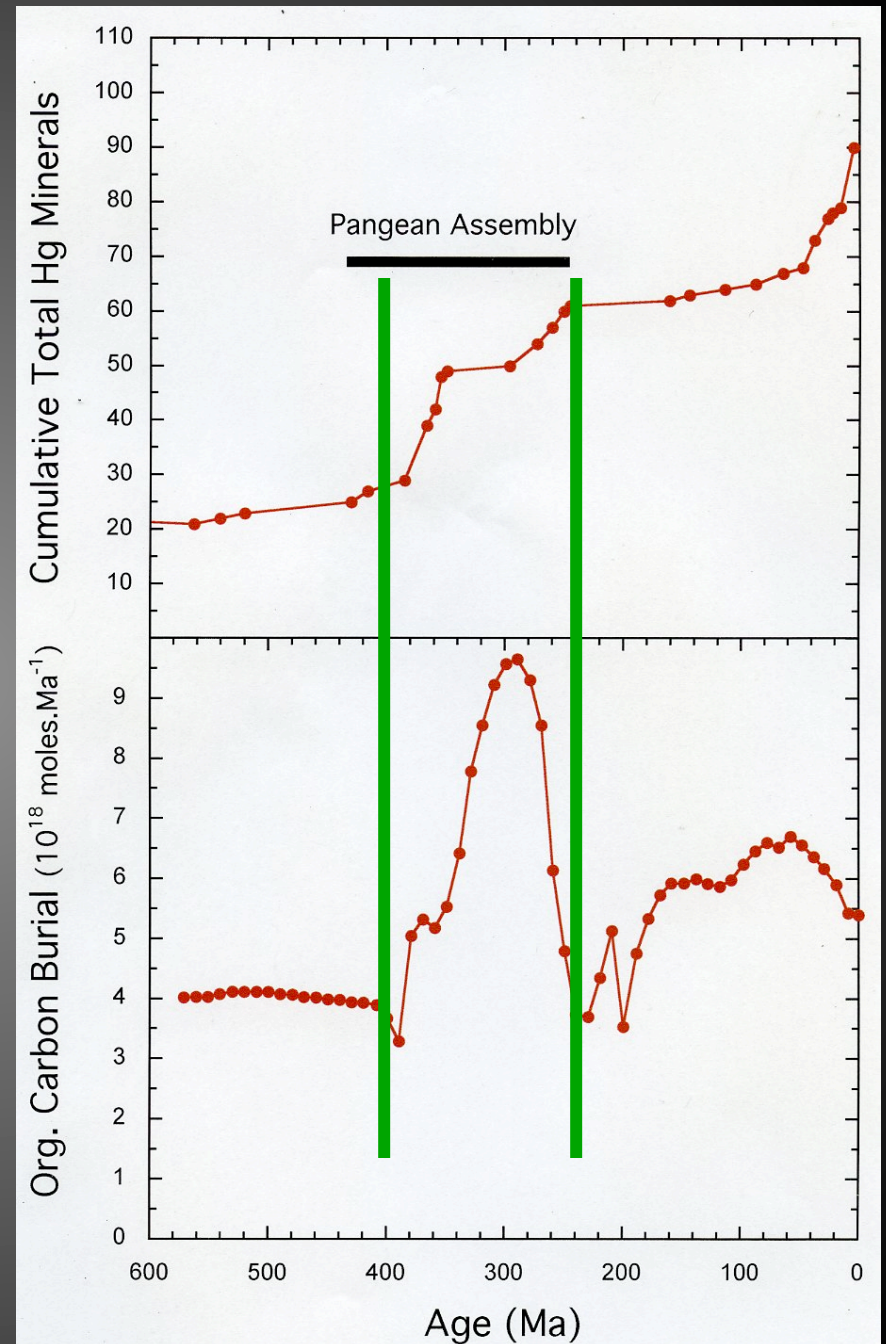
Hg Mineral Evolution

The distribution of mercury (Hg) minerals through the last 450 million years reflects changes in Earth's biosphere.



Hg Mineral Evolution

The distribution of mercury (Hg) minerals through the last 450 million years reflects changes in Earth's biosphere.



Conclusions

With mineral evolution, the science of mineralogy once again assumes its rightful place at the center of the Earth and planetary sciences.



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